

**Hydrothermal evolution of volcanogenic Zn-Cu-Pb deposits, Taylors Brook Showing,
Stirling Belt, southeastern Cape Breton Island, Nova Scotia**

by

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Abstract: Hydrothermal evolution of volcanogenic Zn-Cu-Pb deposits, Taylors Brook showing, Stirling Belt, southeastern Cape Breton Island, Nova Scotia

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The Taylors Brook showing is a suspected volcanogenic massive sulfide (VMS) occurrence (Zn-Pb-Cu-Au-Ag) that occurs within the ~680 Ma Stirling volcanosedimentary-plutonic belt, southeastern Cape Breton Island, Nova Scotia, Canada. In addition to mineralized mudstones, there are 7 associated hydrothermal vein types, characterized using three criteria: vein-filling mineral composition and alteration style, cross cutting relationships and mineralization. Mineralized (Zn-Cu-Pb-Sc-Y-Ag-Te) veins show propylitic and potassic alteration assemblages. Compositional zoning in epidote and a pyrrhotite-smythite replacement reaction indicate that the hydrothermal fluid was experiencing fluctuations in fO_2 and cooling, respectively, rendering the metals less soluble causing their precipitation.

Bulk rock geochemistry indicates mineralization anomalies are associated with host rock Zn-Pb anomalies in pyritic mudstone units and Cu anomalies in type IA (epidote-actinolite-calcite-K-feldspar-quartz-pyrite) veins. Fluid inclusion data from two generations of calcite in later type V veins suggest a $CaCl_2$ -rich hydrothermal brine responsible for metal transport and deposition with bulk salinities ranging between 43-54 wt% total salt (absolute: 33-50 wt% $CaCl_2$ and up to 20 wt % NaCl). The homogenization temperatures for calcite A-hosted inclusions range from 106-306°C (n=21) and in calcite B-hosted inclusions from 212-331°C (n=91). Isochore ranges were plotted using microthermometric data, paired with a po-py equilibrium boundary constraint limits trapping conditions to a maximum of 400°C and 5kbar.

The distinct Ca-Cl-rich brine composition can only be formed naturally under specific conditions. There are four conceivable explanations for this Ca rich fluid: boiling, cation exchange between Na and Ca, evaporation of Ca-rich paleoseawater and dissolution of Ca-evaporites. However, the combination of high salinity and high Ca is only possible via the latter two processes. The Taylors Brook showing has comparable mineralization to the Stirling deposit, and the Duck Pond-Lemarchant deposits of Newfoundland but has an inconsistent fluid chemistry with VMS deposits.

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Table of Contents

Abstract.....	ii
Acknowledgements.....	iii
Table of Contents.....	iv
List of Figures and Appendices.....	v
List of Tables	v
 Chapter	
1.0 Introduction.....	1
2.0 Geological setting.....	2
3.0 Methods.....	6
4.0 Results.....	9
4.1 <i>Mineralization and alteration</i>	9
4.1.1 Type IA veins.....	9
4.1.2 Type IB veins.....	16
4.1.3 Type II.....	18
4.1.4 Type III.....	19
4.1.5 Type IV.....	19
4.1.6 Type V.....	20
4.1.7 Type VI.....	22
4.1.8 Pyritic mudstones.....	22
4.2 <i>Bulk rock analysis</i>	26
4.3 <i>Fluid inclusion study</i>	30
4.3.1 Fluid inclusion petrography.....	30
4.3.2 Microthermometry and salinity.....	35
5.0 Discussion.....	41
5.1 <i>Mineralization and hydrothermal vein paragenesis</i>	41
5.2 <i>Fluid origin in type V veins</i>	42
5.3 <i>Implication for mineral exploration</i>	44
5.4 <i>Comparison to the Stirling deposit and other VMS systems</i>	45
6.0 Conclusions.....	48
7.0 References.....	50
8.0 Appendices.....	55

List of Figures and Appendices

Figure 2.1:	Regional geological map showing the allochthonous terranes of mainland Nova Scotia and Cape Breton Island.....	3
Figure 2.2:	Geological map of southeastern Cape Breton and the Stirling belt.....	5
Figure 2.3:	Local geological map of the Taylors Brook showing.....	7
Figure 4.1:	Paragenetic summary of all vein styles, alteration types and associated ore and gangue mineralogical assemblages.....	10
Figure 4.2:	Photographs of core sampled from drill hole STR-12-02.....	11
Figure 4.3:	SEM-BSE and EDS X-ray maps of type IA veins.....	14
Figure 4.4:	Elemental SEM line scan through zoned epidote.....	15
Figure 4.5:	Cross-cutting vein relationships.....	17
Figure 4.6:	Example of type V veins (comb texture quartz and infilling calcite).....	21
Figure 4.7:	SEM-BSE images of pyrite grains in mudstones.....	23
Figure 4.8:	SEM-BSE and EDS X-ray maps of a pyrite grain enclosed by pyrrhotite..	24
Figure 4.9:	Extensive down hole section of the drill hole STR-12-02.....	27
Figure 4.10:	Bulk rock Zn,Pb,Cu assays and Cu/Zn ratio vs. depth.....	28
Figure 4.11:	Fluid inclusion petrography and example compositional types.....	31
Figure 4.12:	SE images of decrepitate salt mounds on calcite crystal surfaces.....	32
Figure 4.13:	BSE high resolution images of thortveitite within fluid inclusions.....	34
Figure 4.14:	Box and whisker plot of homogenization temperatures.....	36
Figure 4.15:	Trapping pressure-temperature constraints for CaCl ₂ -NaCl-H ₂ O inclusions.....	40
Appendix A:	Relationship of host lithology, alteration style and vein mineralogy to vein type.....	54
Appendix B	Bulk geochemical data obtained via portable XRF.....	55
Appendix C:	Summary of microthermometric data.....	65

List of Tables

Table 4.1:	SEM-EDS analyses of pyrrhotite, smythite and pyrite in textural association in mudstones.....	25
Table 4.2:	Microthermometric and calculated salinity data	39
Table 5.1:	A comparison of the Taylors Brook showing to other VMS systems.....	46 _v

1.0 Introduction

The Taylors Brook showing is a suspected volcanogenic massive sulfide (VMS) occurrence (Zn-Pb-Cu-Au-Ag) that occurs within the ~680 Ma Stirling volcanosedimentary-plutonic belt, southeastern Cape Breton Island, Nova Scotia, Canada. Previous exploration in the region recognized the showing as part of a larger mineralized district containing the polymetallic Stirling VMS deposit (Mindamar Mine, Stirling Mines Ltd.; 1.2 Mt @ 6.4% Zn, 1.5% Pb, 0.74% Cu, 2.2 oz/ton Ag, and 0.03 oz/ton Au) (O'Reilly, 2008).

The genetic relationship between the Stirling deposit and Taylors Brook occurrence is poorly understood. If a VMS the general model for these systems shows that such deposits occur in districts (i.e., multiple sites of mineralization related to hydrothermal systems around a common submarine volcanic spreading center; eg. Noranda: Hannington *et al.*, 2003, Kuroko: Ishihara, 1974, Matagami: Carr *et al.*, 2008, Snow Lake: Bailes *et al.*, 1999 and Flin Flon: Koo *et al.*, 1975). Within such districts, similarities in alteration styles, metal tenor and ore mineralogy, fluid chemistry and associated hydrothermal conditions, and host rock types are expected between deposits. Determining the relationship between the Stirling deposit and the Taylors Brook occurrence first requires a systematic study of such features.

The Stirling deposit was most recently been interpreted by Macdonald and Barr (1993) to have been deposited within, and around, a trough flanked by volcanic centers in an intra-arc setting. A report compiled by Hunter and Gibson (2006; Wallbridge Mining Company Limited exploration project 1998-2005) interpreted the rocks of the Taylors Brook area to be the equivalents of those hosting the Stirling deposit, and represent a

typical submarine volcanogenic setting for massive sulphide deposition hosted within mafic flows, breccias, quartz-feldspar porphyry and mudstone/siltstone (Reid, 2014). The occurrences of VMS mineralization within the Stirling belt have similarities (ie. the association of abundant quartz-carbonate alteration with massive sulphide mineralization) to Proterozoic deposits of the Sudbury area (eg. Errington and Vermillion Zn-Cu-Pb) and the Bergslagen district of Sweden (eg. Garpenberg deposit) (Kontak, 1998).

The main objective of this thesis is to conduct a systematic study of alteration mineralogy and fluid inclusions at the Taylors Brook showing in order to determine conditions of hydrothermal activity, and the relationship of alteration to mineralization. Specifically, optical petrography, scanning electron microscopy (SEM), XRF, and fluid inclusion microscopy and microthermometry are used to characterize the alteration styles, and fluid salinity-temperature pressure characteristics associated with sulfide mineralization and associated hydrothermal events. This will contribute to a genetic and exploration model for suspected VMS mineralization occurrences in the district.

2.0 Geological Setting

The Taylors Brook showing is located in southeastern Cape Breton Island, Saint Esprit. The regional geology in the area of the showing was studied extensively by Barr et al. (1996) and the descriptions here are based on that work. Cape Breton Island is composed of four adjoining terranes: the Blair River inlier, Aspy, Bras d'Or and Avalon (Mira) terranes (Figure 1.1). The site of study lies within the Avalon terrane which docked to Laurentia approximately at 435 Ma. The Avalon Terrane is comprised of five

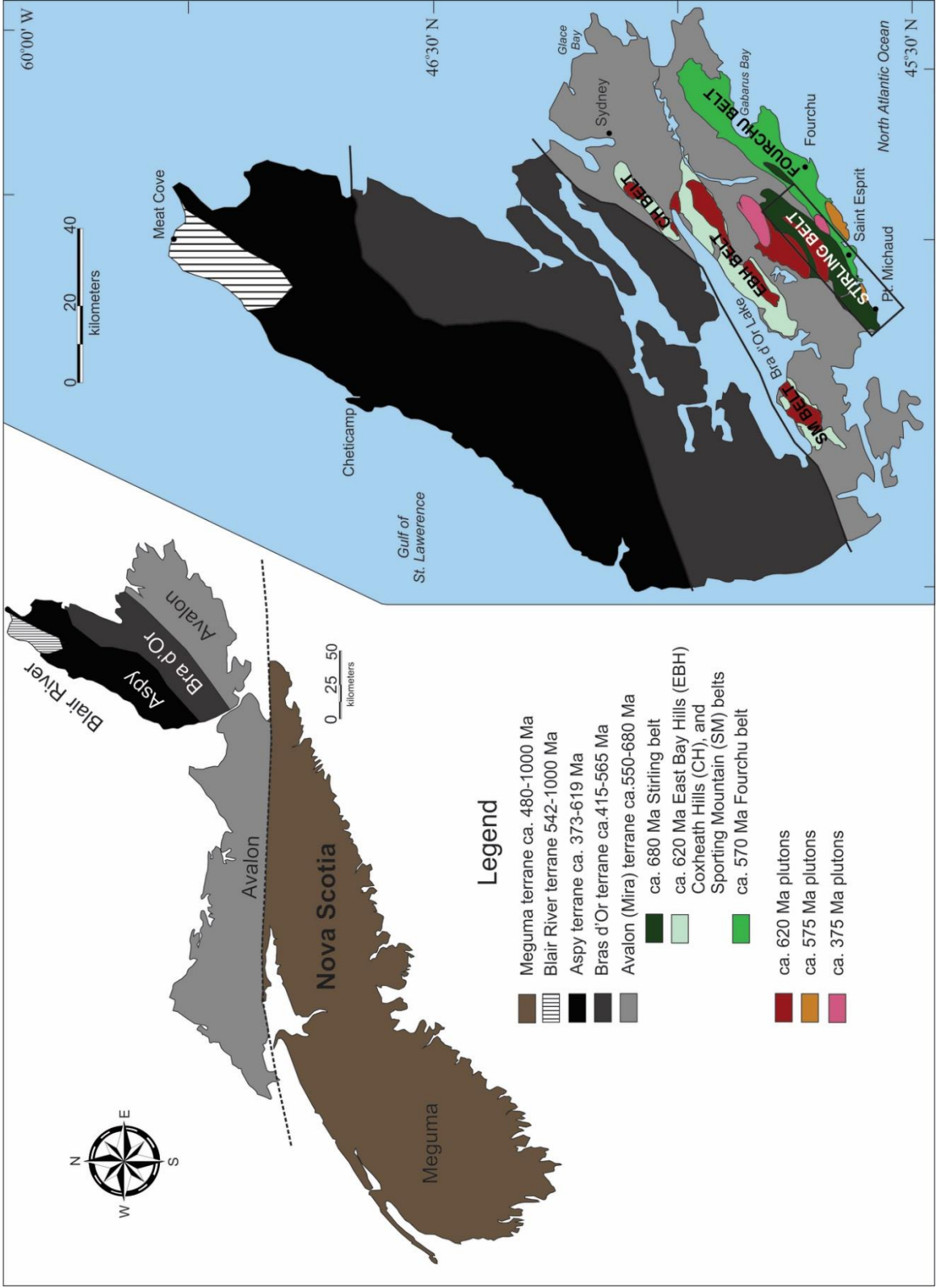


Figure 2.1: previous page. **Regional geological map showing the allochthonous terranes of mainland Nova Scotia and Cape Breton Island.** A) Regional geological map showing the allochthonous terranes of Nova Scotia, specifically those of Cape Breton Island. B) Enlarged view of Cape Breton Island showing the detailed geology of the Avalon terrane. The five north-east trending volcanic-plutonic belts are labelled (compilation of DNR, 2010 & Petts, *et al.*, 2012). The area showing Figure 2.2 is indicated by the black box.

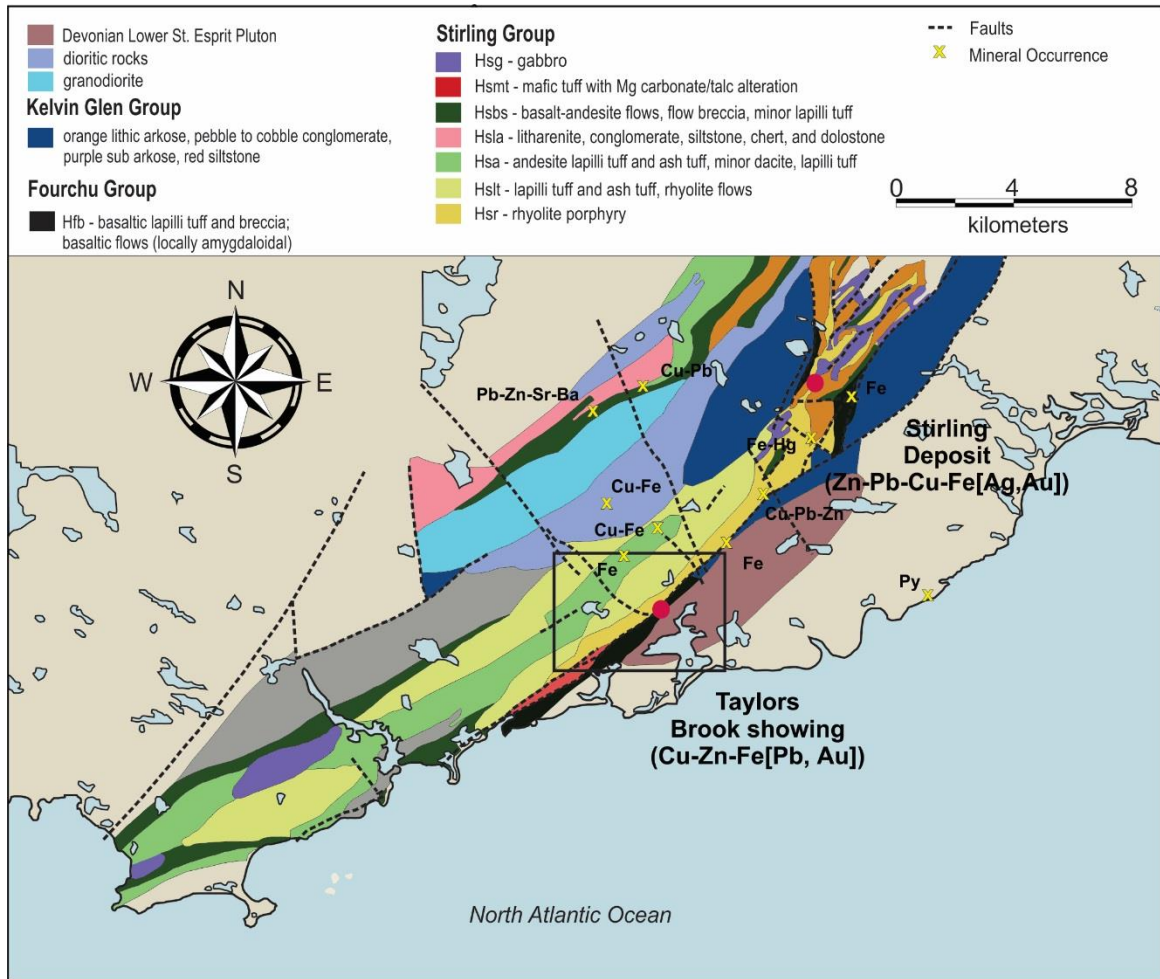


Figure 2.2: Geological map of southeastern Cape Breton and the Stirling belt. Map showing the Stirling belt, Stirling deposit and the Taylors Brook showing (Thundermin Resources Inc., 2012). Mineral occurrences are from the NSDNR mineral occurrence data base. The area of Figure 2.3 is indicated by the black box.

north-east trending, volcanic-plutonic belts. These include the Fourchu or Coastal Belt (575 Ma), Stirling Belt (680 Ma), East Bay Hills Belt (620 Ma), Coxheath Belt (620 Ma) and Sporting Mountain Belt (620 Ma)(Kontak, 1997). The belts were intruded by the Caplin Cove (574 Ma), St. Esprit (375 Ma), Salmon River (375 Ma), Chisholm Brook (620 Ma) and Huntington Mountain (620 Ma) plutons (Figure 2.1).

The Stirling deposit and the Taylors Brook showing are located within the Stirling Belt (Figure 2.1 and Figure 2.2). This belt consists of rocks ranging from felsic to mafic volcanic and volcanoclastics hosting lenses of fine- to coarse-grained clastic sedimentary rocks (Kontak, 1997); (Figure 2.3). The rocks are interpreted to have been deposited within and around a trough-flanked volcanic center in an intra-arc setting (Macdonald *et al.*, 1993). Locally, the Taylors Brook showing is hosted within a variety of lithologies including the Lower St. Esprit granite, basalt/andesite flows, quartz feldspar porphyry, andesitic lapilli tuff and ash. Drill hole STR 12-02, studied here, is surficially located east of the L'Archeveque Fault within the Lower Esprit Granite (Figure 2.3).

3.0 Methods

Extensive re-logging and sampling of core from hole STR-12-02 (Thundermin Resources, 2012) was done in the summer of 2015 in order to characterize and obtain samples for study. During that time, 390 bulk geochemical analyses of drill core sections were made using a X-5000 portable X-ray fluorescence spectrometer (XRF) manufactured by Innov-X Canada, with analyses taken at one meter intervals to obtain major and trace element data.

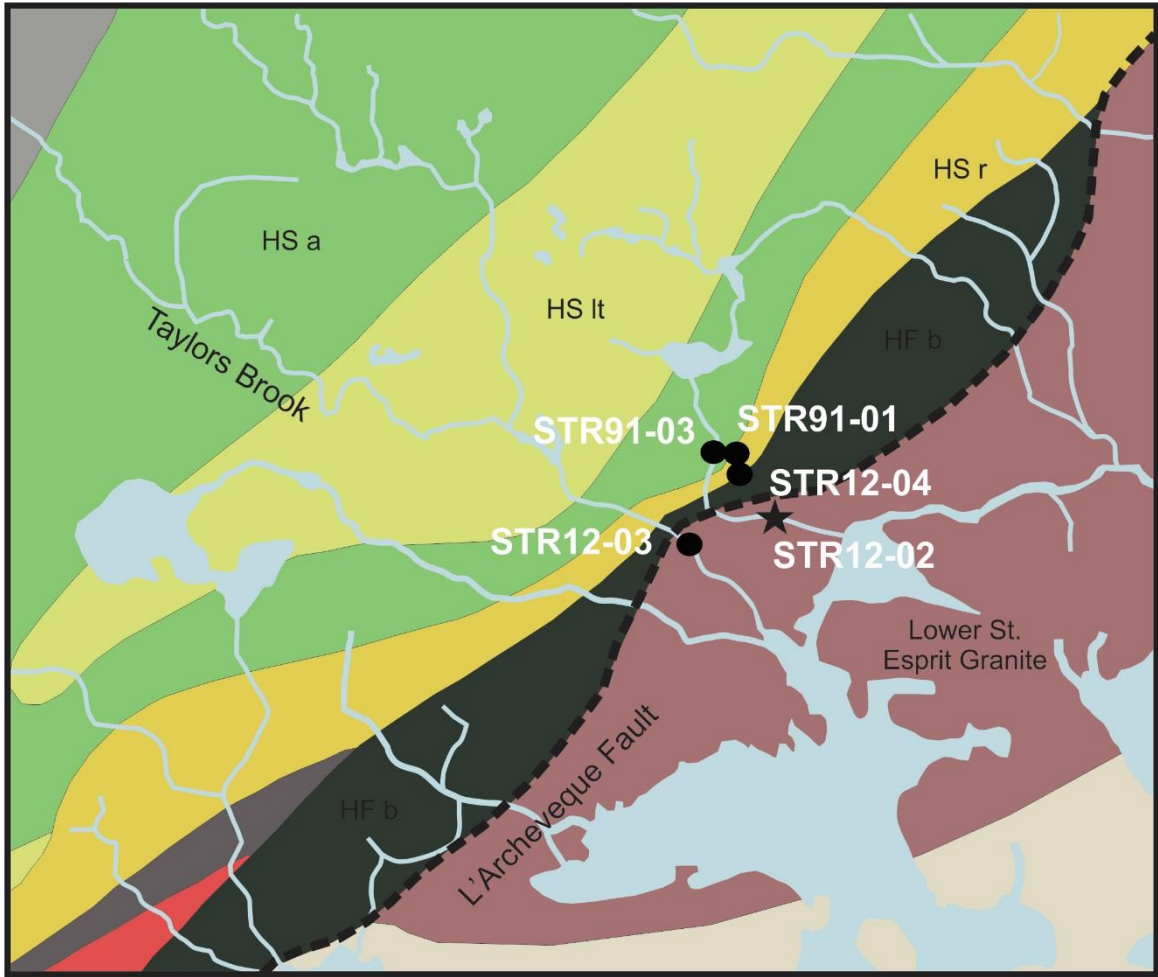


Figure 2.3: Local geological map of the Taylors Brook showing. Map of the Taylors Brook showing after Barr et al. (1996) and Reid (2014). All Taylors Brook showing drill hole collar locations are indicated; Holes STR-91-01 and STR-91-03 drilled by Falconbridge Ltd. in 1991 and STR-12-02, STR-12-03 and STR-12-04 drilled by Thundermin Resources Inc. in 2012. The drill hole sampled and logged for this study is denoted by the black star. Refer to legend in Figure 2.2 for lithological abbreviations.

From the drill hole 40 samples were collected that were representative of cross-cutting vein relationships and alteration and mineralization styles. The samples were cut into blocks and sent to Vancouver Petrographics for thin section preparation (19 polished thin sections for petrographic study and 28 double polished thin sections for fluid inclusion study).

Representative thin sections were chosen for SEM-BSE imaging, SEM-EDS mineral analysis and X-ray mapping. Imaging, analysis and X-ray mapping was done using a TESCAN MIRA-3 LMU Variable Pressure Schottky Field Emission Scanning Electron Microscope (FE-SEM) at Saint Mary's University. An energy dispersive X-ray Oxford INCA X-max 80 mm² EDS system (attached to the SEM) was operated at a working distance of 15-20 mm, a beam current of 40 μ A, and an accelerating voltage of 20 kV. The FE-SEM provided quantitative EDS analyses of minerals as well as a means to identify mineral species, and solids in fluid inclusions.

Microthermometric measurements of fluid inclusions were obtained at Saint Mary's University using a Linkam FTIR 600 heating-freezing stage mounted on an Olympus BX51 microscope. The stage was calibrated using synthetic fluid inclusion standards containing pure CO₂ (melting at -56.6°C) and pure, critical density H₂O (melting at 0°C and homogenizing at 374.1°C). Total uncertainties associated with the microthermometric measurements, based on the reproducibility of measurements conducted on the standards and the measurement (read-out) precision of the controller instrumentation, range from $\pm 2^{\circ}$ to 3°C for temperatures recorded near the extremes of working conditions for the heating-freezing stages (-190° and 560°C), to less than $\pm 0.2^{\circ}\text{C}$ for temperatures recorded near 0°C . Measurements of eutectic (first) ice-melting temperatures (T_e) were obtained to

give insight into the major cation composition of the fluids within the inclusions. Final ice melting temperatures (T_{mice}), antarcticite melting temperatures (T_{mant}), and rarely, halite dissolution temperatures ($T_{mhalite}$) were used variably to calculate inclusion salinities (wt% NaCl and CaCl₂), using the model of Steele-MacInnis et al. (2011). When both hydrohalite and antarcticite melting temperatures were measured in inclusions, CaCl₂ wt% equivalents were estimated graphically using the UNIQUAC model for binary CaCl₂-H₂O systems (Sander *et al.*, 1986). The programs BULK and ISOC (Bakker, 2003) were used to model bulk fluid densities and isochores for fluids in the CaCl₂-NaCl-H₂O and CaCl₂-H₂O systems.

4.0 Results

4.1 Mineralization and alteration

Samples were characterized using three criteria: vein-filling mineral composition and alteration style (ie. vein and wall rock assemblages), cross cutting relationships (ie. relative timing of veining styles) and mineralization (ie. presence/absence of Zn-Cu-Pb-Fe sulphides). Petrographic studies of the veins allowed for a division into 7 vein types (Figure 4.1). The relative order of formation of type IA, IB, II, III, IV, V and VI veins was determined by examination of cross cutting relationships in hand sample and thin section (oldest to youngest).

4.1.1 Type IA veins

Type IA veins show a continuous potassic alteration rind (~1 cm wide) and are rimmed with epidote. Vein minerals are epidote, quartz, carbonate, actinolite, K-feldspar and trace sulphides (Figure 4.2 B, D and I). Type 1A veins range from 0.5-

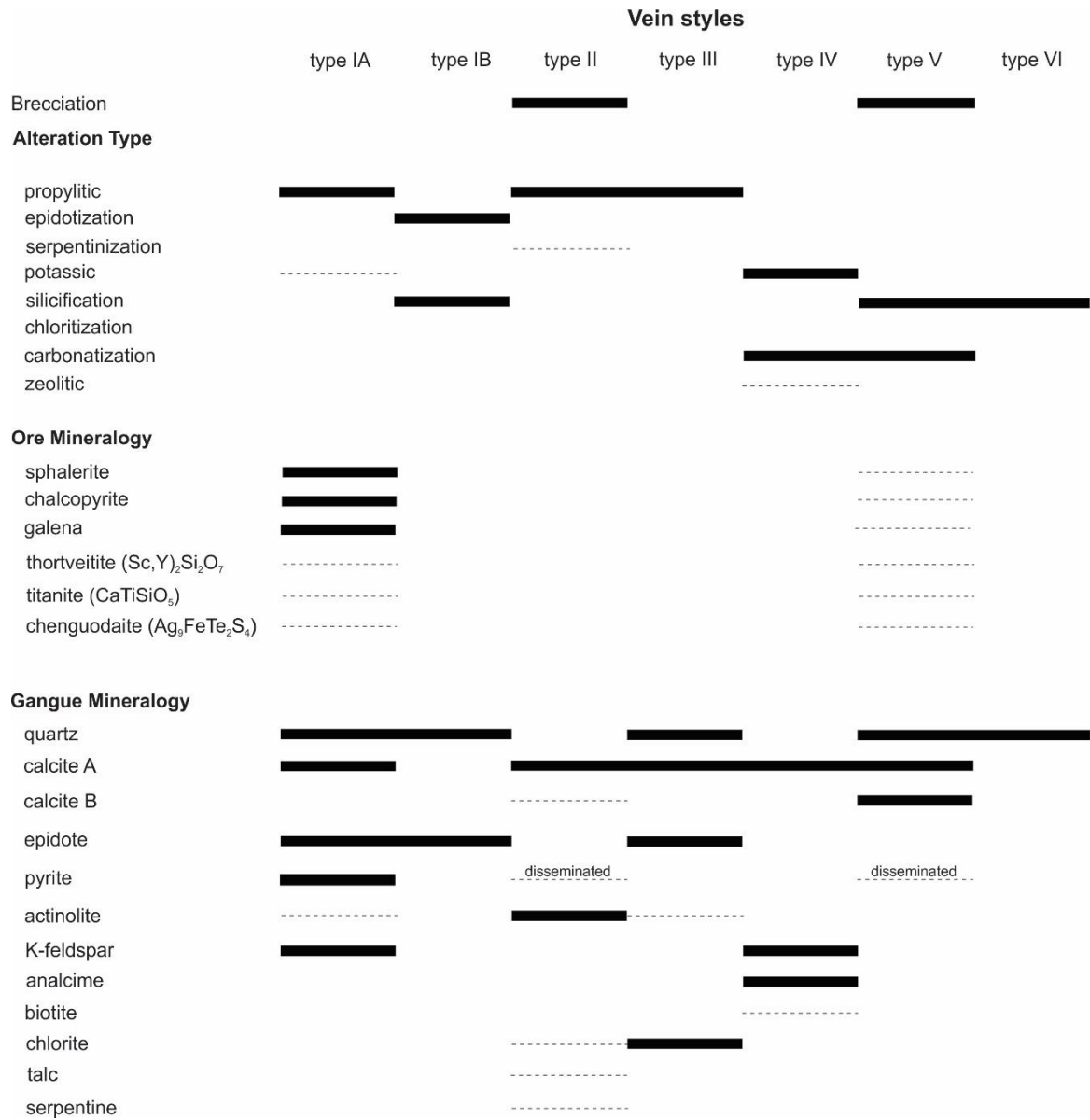


Figure 4.1: Paragenetic summary of all vein styles, alteration types and associated ore and gangue mineralogical assemblages.

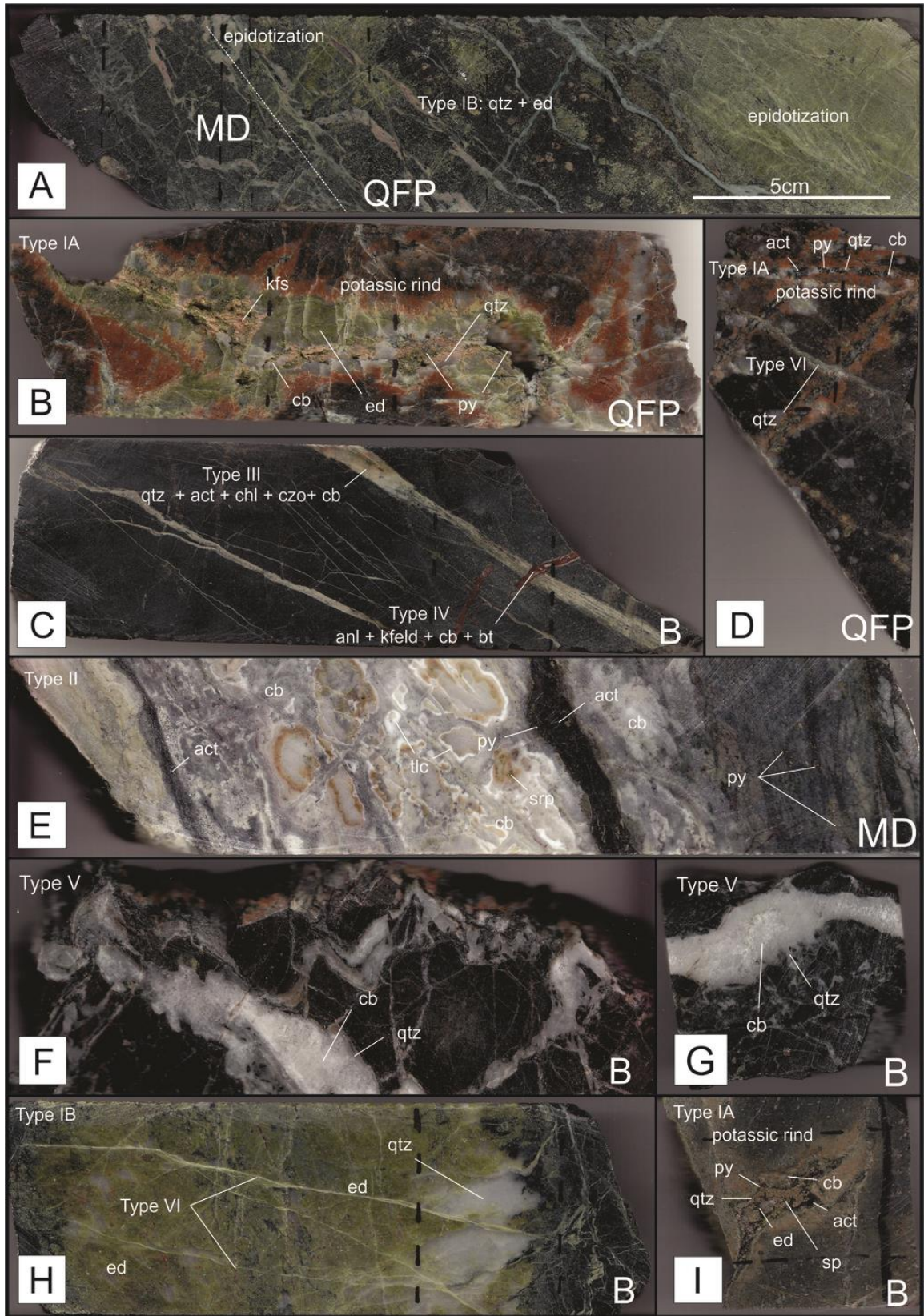


Figure 4.2: previous page. **Photographs of core segments sampled from drill hole STR-12-02.** A) Sample 412.32; an image of the Quartz-Feldspar Porphyry (QFP) cut by a mafic dyke (left hand corner). Both lithologies were highly epidotized, followed by an overprinting of fractured controlled quartz-carbonate veining. Epidotization and veining postdate emplacement of the mafic dyke indicating sulphide mineralization must be younger than the mafic dyke. B) Sample 220.18; an image of the QFP containing a large (1-3cm), type IA (ed-qtz-cb-kfs) vein. This type IA vein contains minor sulphides (py), most likely indicating a higher temperature. It also displays a thick potassic rind characteristic of the type IA fluids. C) Sample 108.9; an image of basalt demonstrating exceptional cross cutting relationships between type III (qtz-act-chl-czo-cb) and type IV (anl-kfs-cb-bt) veins. The image shows that type III veins must have been in place before type IV. The type IV vein is further separated along a mini fault plane. D) Sample 284.15; an image of the QFP showing the cross cutting relationship between type IA (ed-qtz-cb-kfs-act) and type VI (qtz) veins. The image clearly displays sulphide rich type IA veins being crosscut by a later type IV vein. E) Sample 155.65; Example of a large type II (act-cb-tlc-srp-py) vein. F) Sample 106.17; an image of brecciated basalt containing a barren type II (qtz-cb) vein. G) Sample 95.1; Very similar to F) an image of brecciated basalt containing a type II (qtz-cb) vein. H) Sample 406.0; an image of basalt containing a large (10cm) type IB (ed-qtz) vein crosscut by a fracture controlled type VI. I) Sample 113.30; an image of pyritic mudstone featuring a type IA vein sulphide rich (qtz-ed-act-sp-py) vein.

Mineral abbreviations: act=actinolite, anl=analcime, bt=biotite, cb=carbonate, chl=chlorite, czo=clinozoisite, ed=epidote, kfs=k-feldspar, py=pyrite, qtz=quartz, srp=serpentine, sp=sphalerite, tlc=talc. Scale applies to all frames.

3 cm in width and have comb textured margins with a vuggy infilling. Epidote (70 vol. %) is massive, with characteristic second order interference colors in CPL and a green-yellow color in PPL, suggesting high Fe content. Two types of carbonate (A and B) are present within some veins but only carbonate A (12 vol. %) is found within type IA veins. It occurs as subhedral grains between 0.3 and 2 mm in diameter. Carbonate A has a crystalline structure with distinctive twinning planes and shows a milky appearance. The grains display fifth order interference color in CPL and are colourless-light grey in PPL. Quartz (10 vol. %) occurs as subhedral grains between 0.1 and 2 mm in diameter. The grains display a grey first order interference color in CPL and are transparent in PPL. Actinolite (2 vol. %) is commonly present as small (0.5 mm), radiating acicular grains. The actinolite displays characteristic yellow-blue second order interference colors in cross-nicols and a pale yellowish color in plane polarized light (PPL). Chalcopyrite (1 vol. %) and sphalerite (1 vol. %) are found most commonly where massive pyrite (3 vol. %) occurs. Trace minerals found within type IA veins are thortveitite ($[\text{Sc},\text{Y}]_2\text{Si}_2\text{O}_7$), titanite (CaTiSiO_5) and chenguodaite ($\text{Ag}_9\text{FeTe}_2\text{S}_4$).

Figure 4.3 shows a type IA vein (SEM-BSE) with quartz as subhedral grains displaying an open-space filling texture intergrown with epidote. Massive epidote is commonly compositionally zoned surrounding sulphide mineralization. A line scan through an epidote grain showing zoning in BSE images confirmed variations of epidote-clinozoisite solid solution composition represented by an increase in Fe and a decrease in Al with dark to light BSE grayscale variations (Figure 4.4).

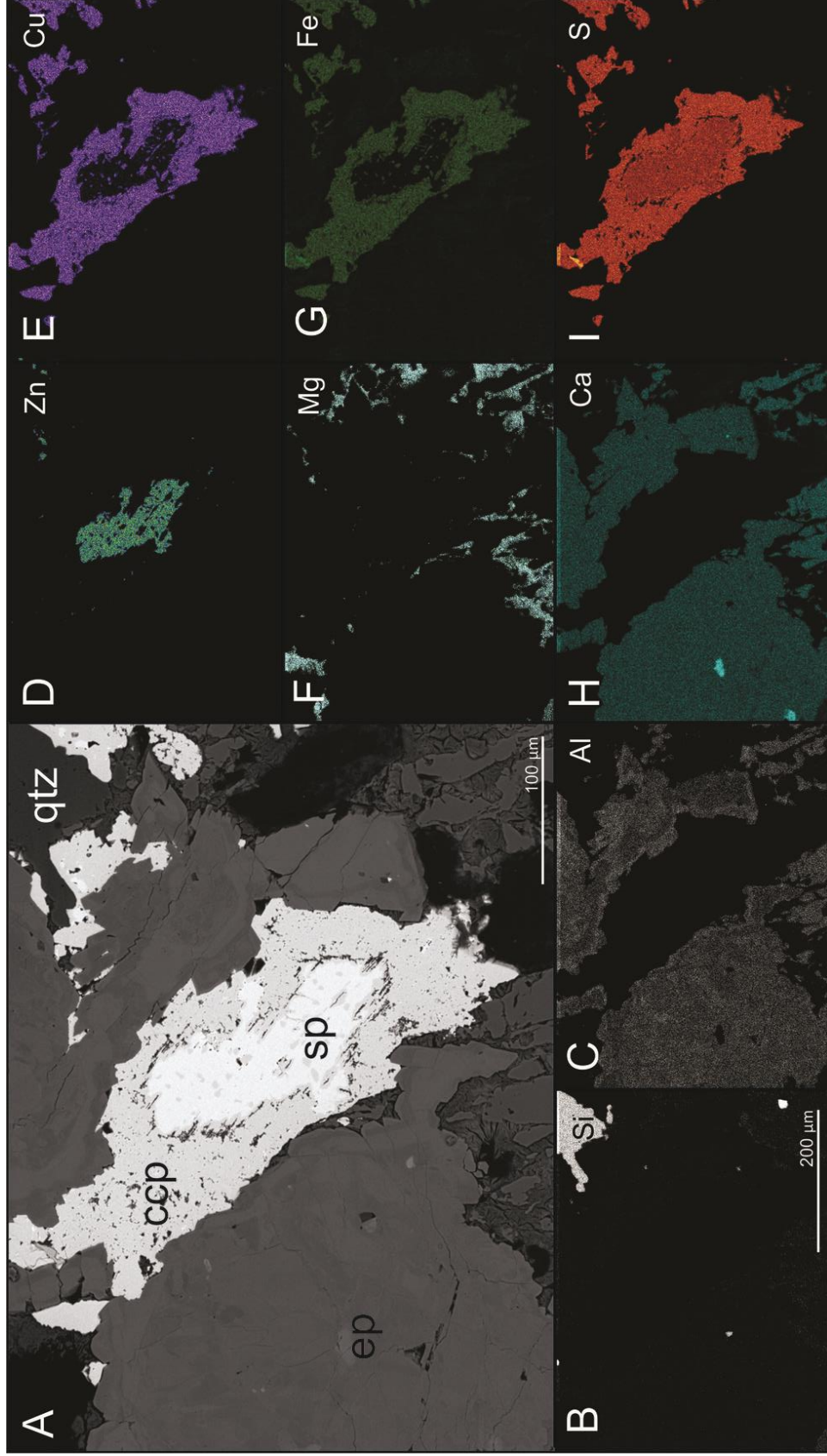


Figure 4.3: BSE and X-ray maps of mineralized type IA vein. A) SEM-BSE of mineralized type IA vein showing a sphalerite (sp) grain enclosed by chalcopyrite (ccp) surrounded by epidote (ep). SEM-EDS X-ray maps (B-I) are highlighting specific element distributions (higher concentration = brighter). B) Showing the distribution of Si. C) Showing the distribution of Al. D) Showing the distribution of Zn E) Showing the distribution of Cu. F) Showing the distribution of Mg. G) Showing the distribution of Fe. H) Showing the distribution of Ca and I) showing the distribution of S.

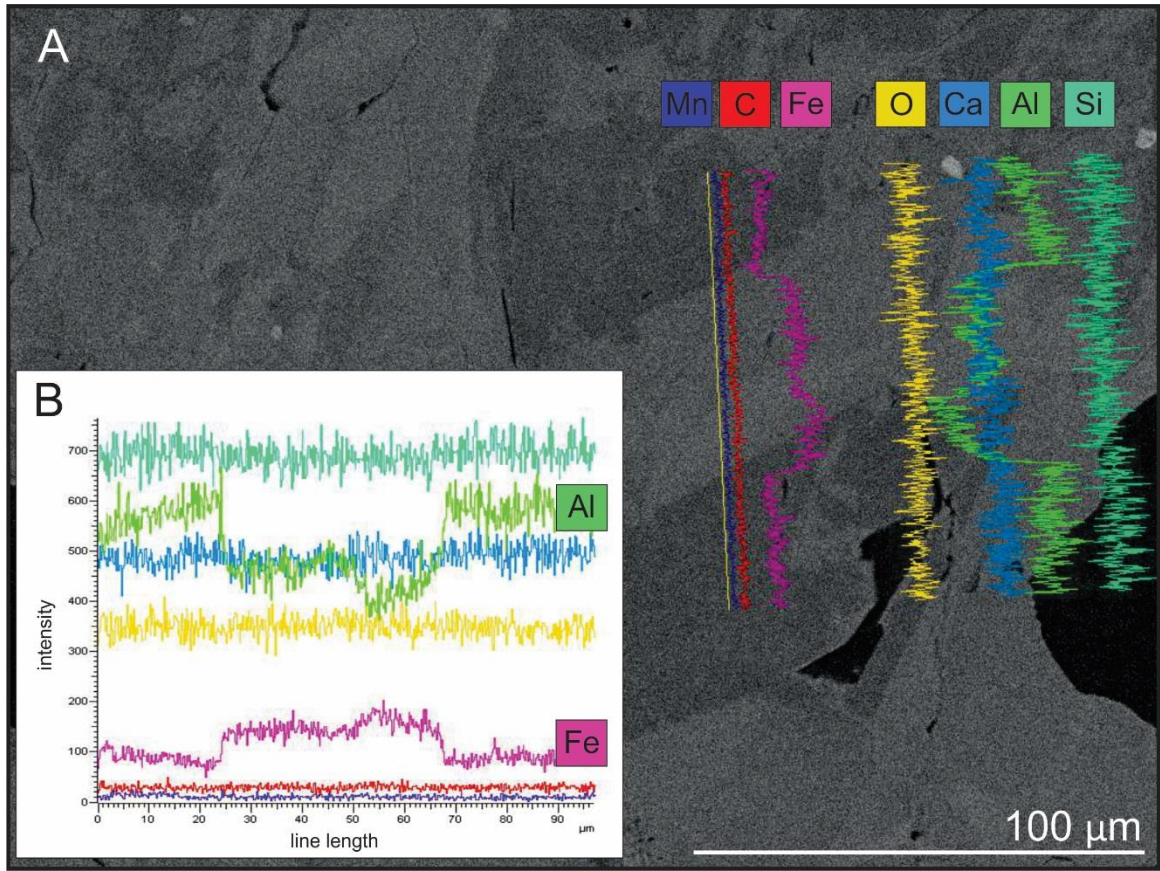


Figure 4.4: Element SEM line scan through a section of zoned epidote shows the variation in solid-solution from iron-rich (magenta) to aluminum-rich (green). A) SEM-BSE image showing the direct placement of the analysis line (line on left) and intensity of elements (left to right) Mg, C, Fe, O, Ca, Al and Si along that line. B) The analysis line transposed on a graph (line length vs. intensity).

Chalcopyrite displays an open-space filling texture with epidote. Sphalerite shows selective replacement texture with chalcopyrite and also chalcopyrite “disease”. Galena although not present in figure 4.3 is most commonly found as traces rimming chalcopyrite, pyrite or sphalerite. Titanite and chenguodaite are both found throughout type IA veins and pyritic mudstones as microscopic (~5 µm) inclusions enclosed by most commonly pyrite but also chalcopyrite and sphalerite. These textures are representative of all type IA veins samples within this sample set.

Type IA veins have irregular vein margins and show alteration haloes of alkali feldspar and epidote. Type IA veins display propylitic-potassic alteration assemblage (epidote-actinolite-calcite-K-feldspar-quartz-pyrite) and are commonly mineralized. Figure 4.5B shows an example of this vein type, illustrating that it postdates IB veins.

4.1.2 Type IB veins

These veins contain massive epidote and quartz, and range from 2 mm to 10 cm in width (Figure 4.2H). The massive epidote (90 vol. %) is similar in its optical and textural characteristics to the epidote in type IA veins. The massive quartz (10 vol. %) displays a grey first order interference color in CPL and is transparent in PPL. Type IB veins have irregular vein margins and alteration halos of epidote. They show an epidotization-silicification alteration assemblages (epidote-quartz).

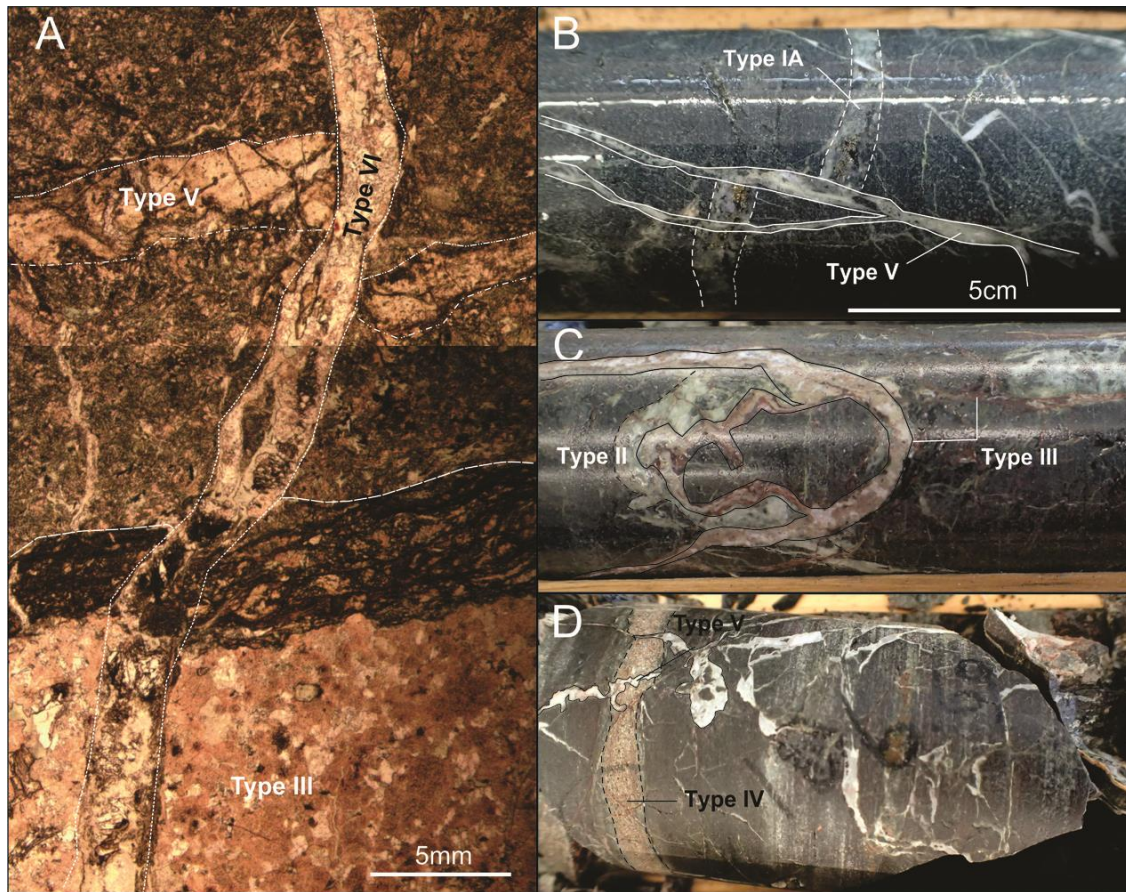


Figure 4.5: Cross-cutting vein relationships. Hatched lines are added to all images to help illustrate cross cutting relationships. Scale applies to all frames A) A cross-polarized transmitted light photomicrograph, of a type V vein cross-cut by a later type VI vein. A type III cross-cut by the type VI is also visible. B) A photograph of a type IA vein (note no potassic rind due to mafic host lithology) cross-cut by a later type IB. C) Photograph of a type II cross-cut by a later type III. D) Photograph of a type IV being cross cut by a type V.

4.1.3 Type II veins

Type II veins are rimmed with actinolite and are infilled with two types of carbonate (A and B), disseminated sulphides, and centrally-occurring, zoned, globular masses of serpentine and talc related to the hydrothermal alteration of orthopyroxene and olivine (Figure 4.2E). Type II veins range from 1 to 15cm in width and have a massive appearance. The majority of the veins are comprised of the two generations of massive carbonate. Carbonate A (80 vol. %) has a massive habit and a milky appearance. Carbonate B (5 vol. %) has a very crystalline appearance with distinct grain boundaries, is transparent and has very distinctive twinning planes. The grains display fifth order interference colors in CPL and are colourless-light grey in PPL. Actinolite (10 vol. %) appears to be massive around the vein boundaries. However, there are some small (0.3 mm) acicular radiating grains. The actinolite displays similar optical characteristics to that in type IA veins. Chlorite (2 vol. %) occurs as radiating grains approximately 0.2-0.5 mm in diameter. The grains show characteristic Berlin blue in CPL and are yellowish-grey in PPL. Talc (1 vol. %), serpentine (1 vol. %) and disseminated pyrite (1 vol. %) are minor phases. Type II veins have irregular vein margins and brecciated host rock clasts are commonly located within the vein. Type II veins display serpentization-propylitic alteration assemblages (serpentine-talc-chlorite-actinolite-calcite) and are rarely mineralized. Figure 4.5C shows an example of this vein type, illustrating that it predates type III veins.

4.1.4 Type III veins

These veins contain quartz, carbonate, chlorite, actinolite and epidote (Figure 4.2C). Quartz (30 vol. %) occurs as subhedral grains 0.1-0.8 mm in diameter. The grains have the same optical characteristics as seen in type II veins. Carbonate A (40 vol. %) has the same optical and textural characteristics as in type IA veins. Epidote (26 vol. %) occurs as subhedral grains 0.2-1 mm in diameter. The grains show characteristic second order interference colors in CPL and a yellowish-green color in PPL indicating a high Al content. Chlorite (2 vol. %) occurs as radiating grains approximately 0.5mm in diameter. The grains show similar optical characteristics as type IB vein chlorite. Actinolite (2 vol. %) occurs as small (0.2-5mm) radiating grains. The actinolite displays similar optical characteristics to actinolite in type IA veins. Type III veins have defined vein margins and they contain a propylitic alteration assemblage (chlorite-epidote-actinolite-calcite). Figure 4.5C shows an example of this vein type, illustrating that they postdate type II veins.

4.1.5 Type IV veins

Type IV veins are composed of carbonate, K-feldspar, analcime and biotite (Figure 4.2C and 4.5D). Type IV veins range from 0.5-3cm in width and appear as linear veins. Carbonate A (10 vol. %) occurs as subhedral grains between 0.1-0.5 mm in diameter. The grains have the same optical characteristics as in type IA veins. K-feldspar (60 vol. %) occurs as euhedral grains 0.1-0.3 mm in length. All grains of K-feldspar show good calrsbad twinning, display a grey first-order interference color in CPL and are transparent in PPL. Analcime (15 vol. %) occurs as subhedral-euhedral grains 0.3-0.8 mm in diameter. The grains are nearly isotropic in CPL and

are transparent in PPL, showing characteristic twinning. Biotite (15 vol. %) occurs as euhedral grains 0.1-0.3 mm in length. The grains display pinkish-green fourth order interference colors in CPL and are brown and extremely pleochroic in PPL. Type IV veins have defined vein margins and show a potassic-carbonatization-zeolitic assemblage (biotite-K-feldspar-calcite-analcime).

4.1.6 Type V veins

These veins were the focus of *fluid inclusion analysis*. The veins contain bladed, comb textured, clear quartz crystals that radiate out from vein walls creating vugs infilled with carbonate A and B (former open spaces) (Figure 4.2 F and G). Type V veins range from 1 to 5cm in width and form irregular networks surrounding brecciate host fragments. Quartz (40 vol. %) occurs as large euhedral grains between 2-6 mm in length. Carbonate A (30 vol. %) occurs with the same optical and textural characteristics as in type IA veins. Carbonate B (10 vol. %) occurs as euhedral grains between 1-3mm in width. The grains have the same optical and textural characteristics as in type II veins. Type V veins comprise breccias containing host fragments but show well defined vein margins. They show a silicification-carbonatization assemblage (quartz-calcite) and are rarely mineralized. Figure 4.5D shows an example of this vein type, illustrating that they postdate of type IV veins.

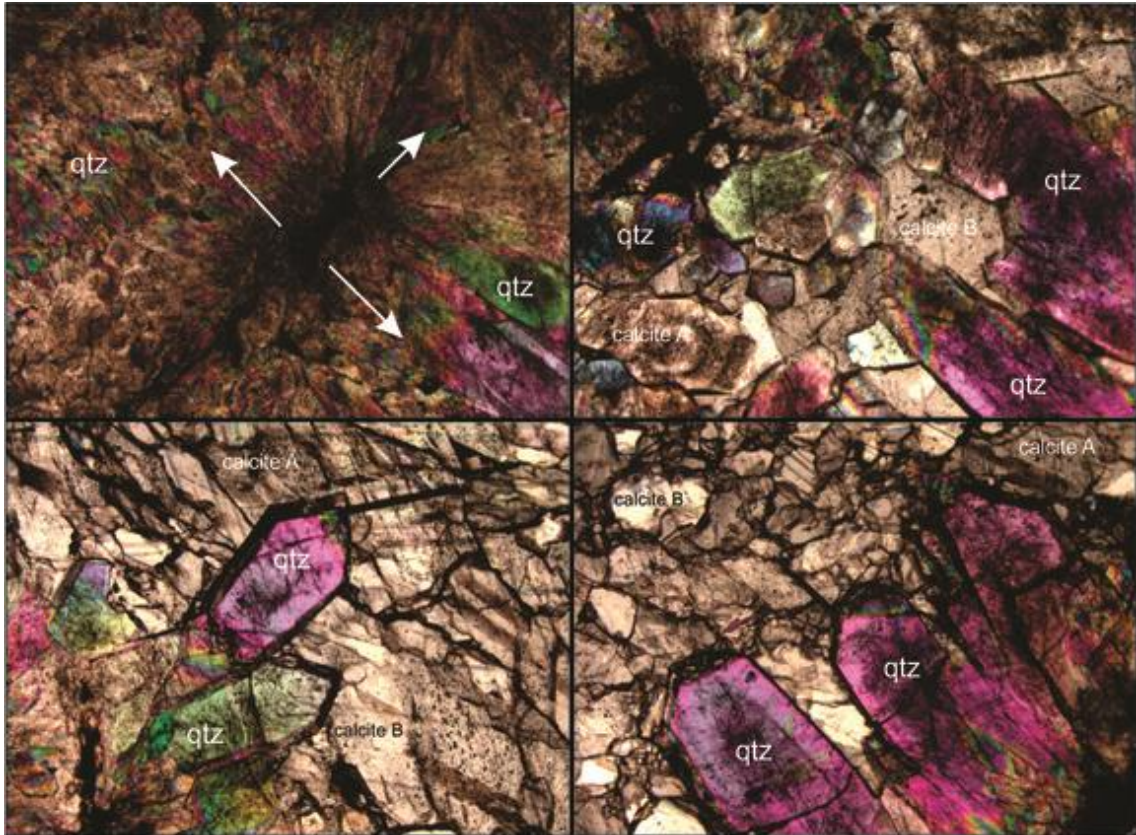


Figure 4.6: Example of type V veins (comb texture quartz and infilling calcite). Calcite A and B in type V veins were the focus of *fluid inclusion analysis*. Arrows are showing growth patterns in quartz away from a nucleation point. Note that the interference colors are anomalous due to thickness (100µm) of the fluid inclusion sections giving the quartz the appearance of epidote.

4.1.7 Type VI veins

The youngest vein styles contain quartz only. Type VI veins are 3 to 5mm in width. Quartz occurs as subhedral grains 0.1-0.8 mm in diameter. The grains have the same optical characteristics of type II but show a more polycrystalline texture. Type VI veins represent a silicification assemblage and have sharp vein margins. Figures 4.2A-C and Figure 4.5A show an examples of this vein type and its cross cutting relationships with other vein styles.

4.1.8 Pyritic mudstones

In the mudstones, a distinct compositional and textural boundary was seen between Fe-S phases (Figure 4.8). SEM-EDS analyses along edges of pyrite grains in contact with the other Fe-sulphide (Figure 4.8) were obtained to determine which occurrences were pyrrhotite rather than smythite. Smythite ($[\text{Fe}, \text{Ni}]_9\text{S}_{11}$) is a more Fe-rich variation of pyrite. Smythite occurs as anhedral grains that are extensively fragmented and skeletal; these enclose pyrite which occurs as cubic grains (Figure 4.8). Sphalerite, galena and chalcopyrite are found either infilling smythite fragments or rimming pyrite, pyrrhotite or smythite (Figure 4.7). This suggests Zn-Cu-Pb mineralization postdates this pyrrhotite-smythite replacement reaction. There were many examples of this replacement reaction wherever pyrite was found and one example with pyrrhotite enclosing pyrite (Figure 4.9). The compositional variation between the smythite, pyrite and pyrrhotite was confirmed with SEM-BSE and EDS X-ray mapping (Table 4.1).

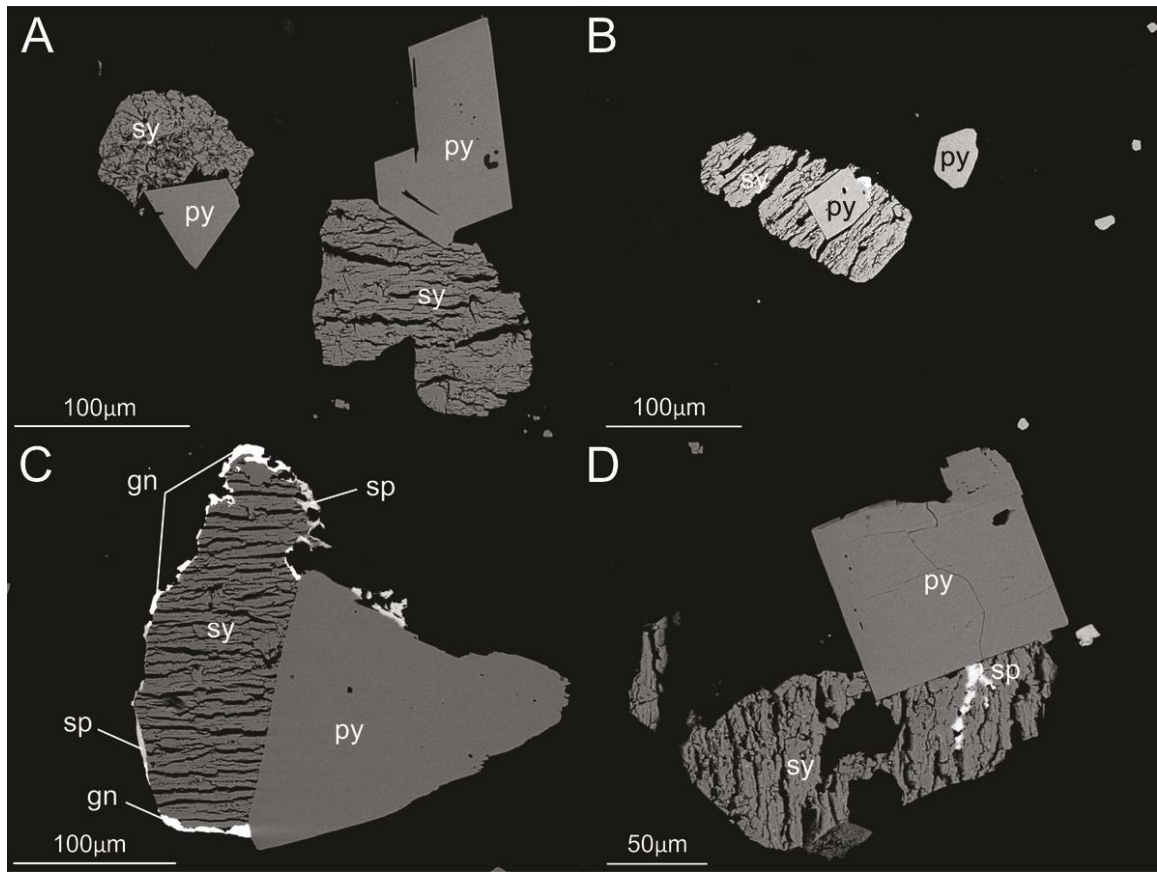


Figure 4.7: SEM-BSE images of pyrite grains in mudstones. A) Two fragmented and skeletal smythite grains with their adjoining cubic pyrite grains. B) A cubic pyrite grain enclosed within skeletal smythite. C) A fragmented and skeletal smythite grain and an adjoining pyrite, rimmed with sphalerite and galena. D) A smythite grain and an adjoining cubic pyrite with sphalerite forming from the boundary edge through the skeletal smythite.

Mineral abbreviations: gn=galena, py=pyrite and sp=sphalerite, sy=smythite

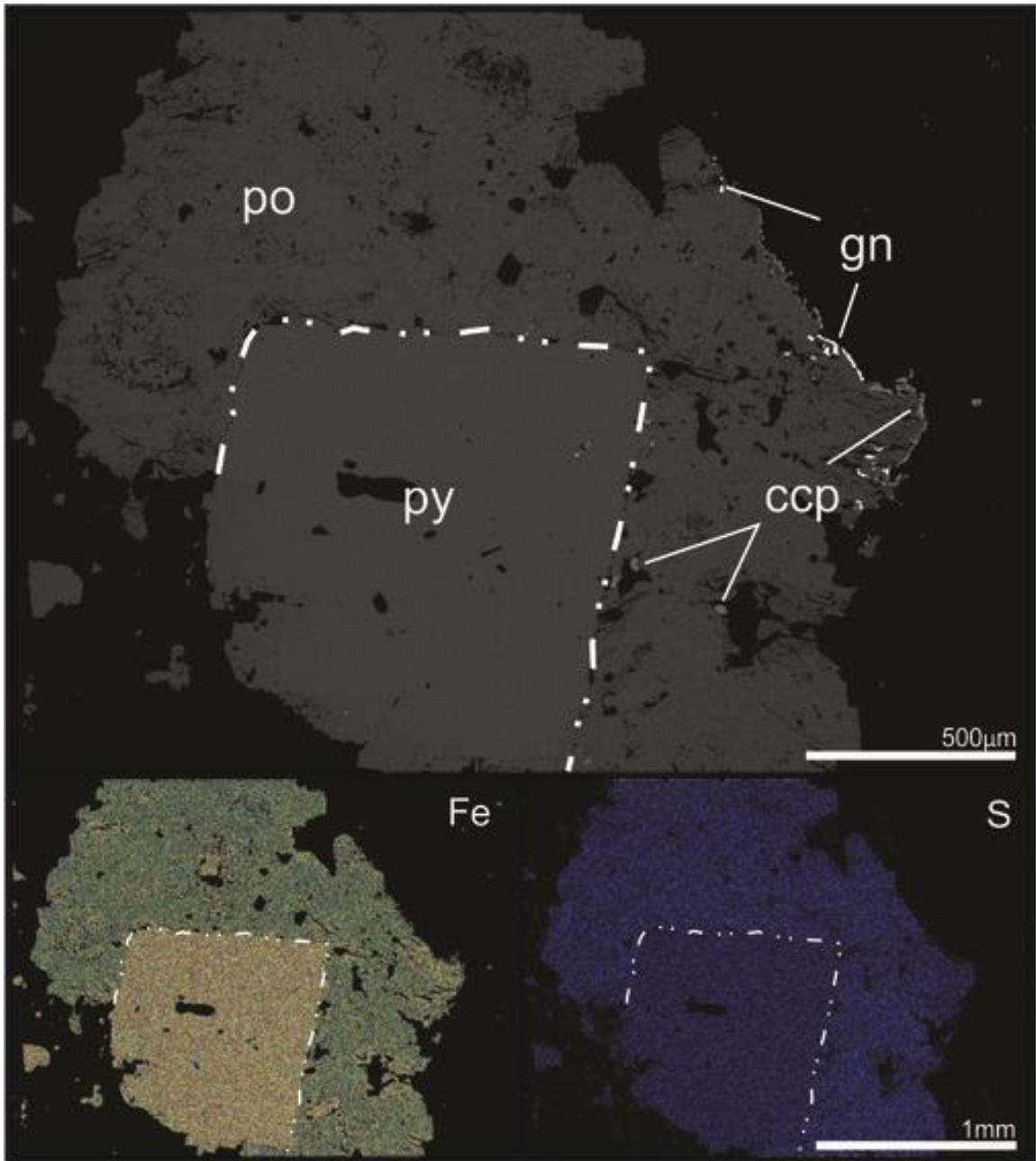


Figure 4.8: SEM-BSE and EDS X-ray maps of a pyrite grain enclosed by pyrrhotite. The added hatched white line illustrates grain boundary. A) SEM-BSE image of a large pyrite grain enclosed by pyrrhotite. Pyrrhotite is rimmed with galena and chalcopyrite. SEM-EDS X-ray maps B) highlighting an increase of S within the cubic pyrite grain and C) highlighting an increase of Fe within the fragmental (higher concentration=brighter).

Mineral abbreviations: ccp=chalcopyrite, gn= galena, py=pyrite, po= pyrrhotite, sp=sphalerite

Each intergrowth most likely began as pyrite-pyrrhotite, but either upon cooling or an influx of S into the system, pyrrhotite converted to smythite. Equilibrium between pyrite and pyrrhotite only occurs within a specific fS_2 and temperature range (Toulmin and Barton, 1964). Thus, analysis of pyrrhotite in equilibrium with pyrite allows the temperature and fS_2 of the mineralizing fluid to be estimated using the calibration of Toulmin and Barton (1964). The analysis provided a maximum mineralizing temperature constraint of 400°C.

4.2 Bulk rock analyses

Analyses of drill hole STR-12-02 were taken at one meter intervals with a portable XRF to obtain major and trace element concentrations at different depths, allowing an examination of the relationship between mineralization and whole rock chemistry. The first mineralized anomaly (Zn-Cu-S) was observed at 113.3 m depth hosted within a type IA mineralized vein in basalt. At 150.2 m depth a (Zn-Pb-Cu-S) anomaly was seen within pyritic mudstone (Figure 4.6). Anomalies at 227.2 m, 228.4 m, 247.19 m, 252 m, 256.2 m and 259.07 m (Zn-Pb-S) correspond to type IA mineralized veins within the QFP (Figure 4.6). At depths between 302.26 to 308.1 m anomalies (Zn-Pb-S) were again observed in pyritic mudstone (Figure 4.6) and a (Cu-Fe-Pb-S) anomaly was observed at 415.30 m (Cu-Fe-Pb-S) as a type IA vein hosted in mafic volcanoclastic (Figure 4.6). All geochemical data is summarized in the appendix. It is clear that major mineralization anomalies are associated with pyritic mudstone units and type IA veins.

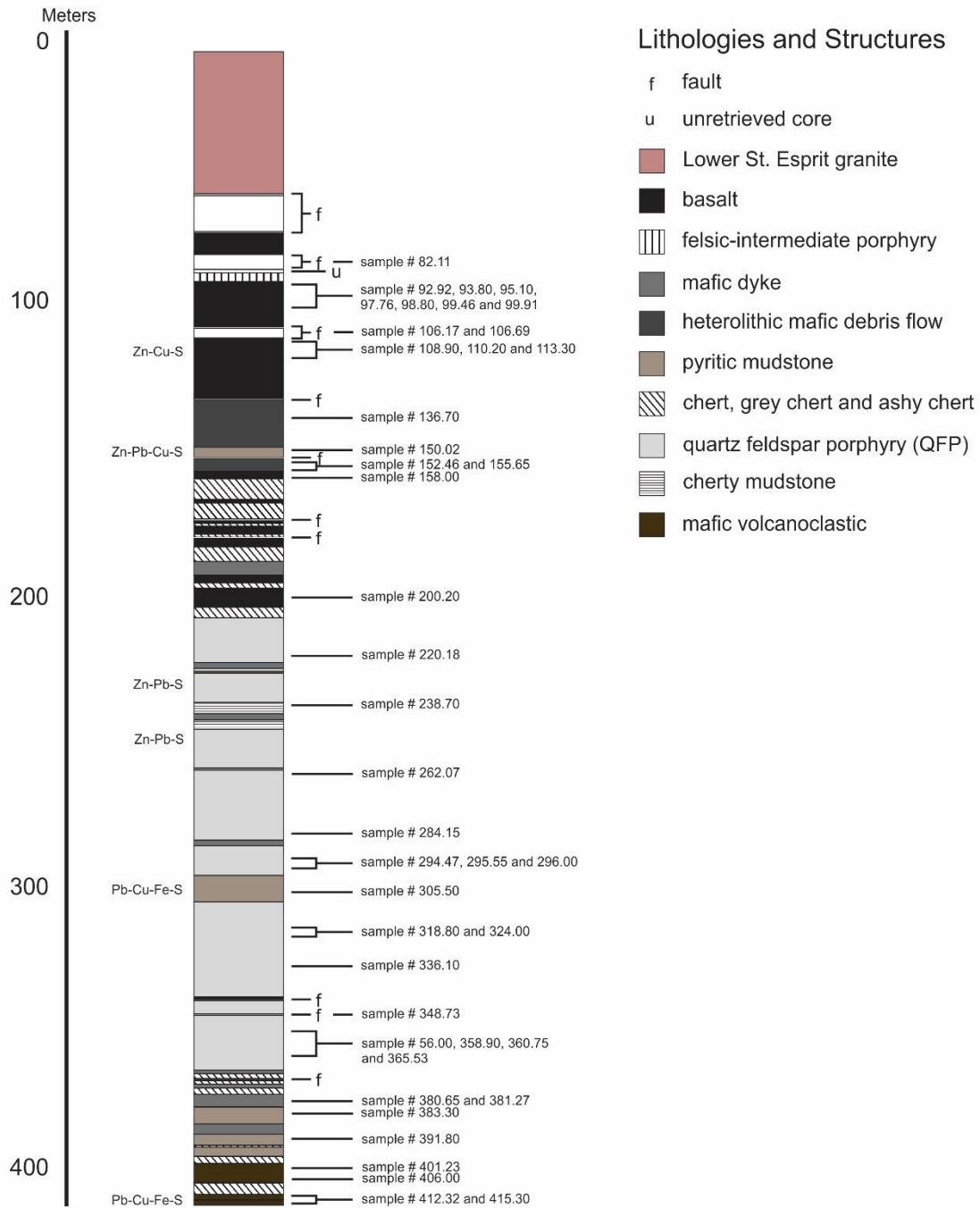


Figure 4.9: Extensive down-hole section of the drill hole STR-12-02. The section shows the host lithology relationships, sample locations and mineralization spikes. Detailed core log and XRF bulk rock analysis presented in Appendix.

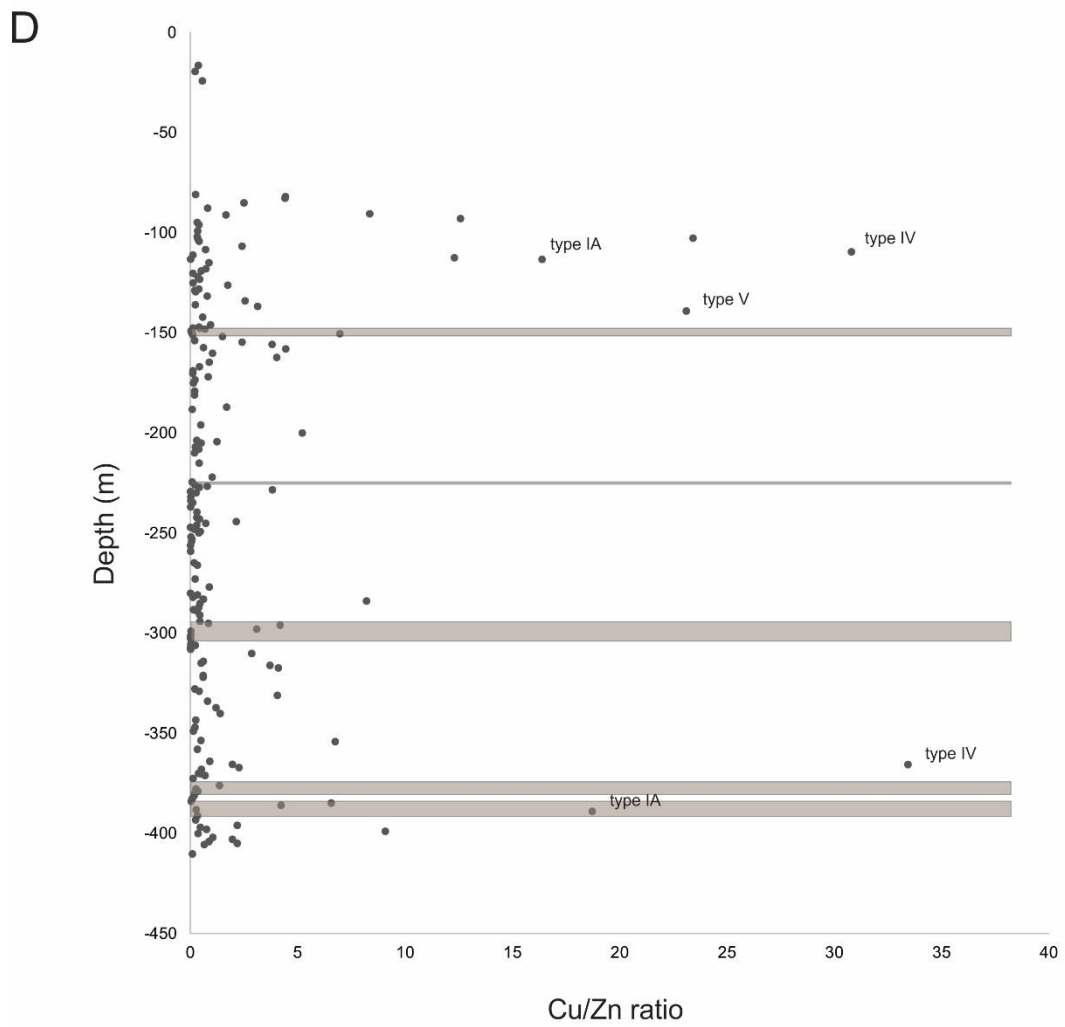
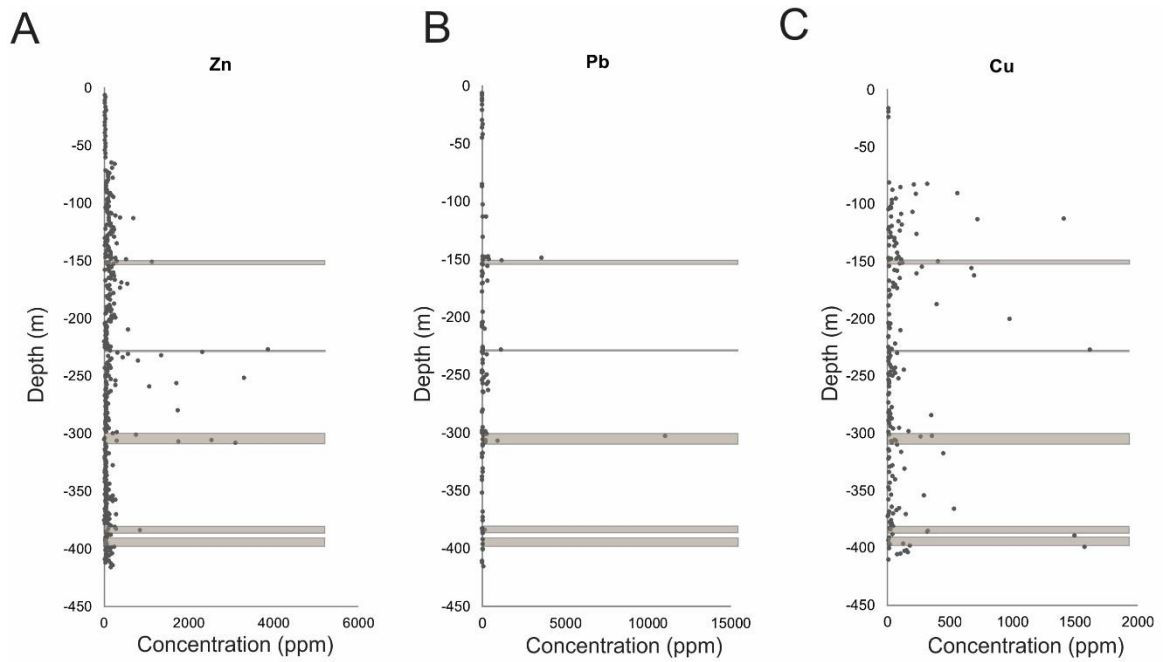


Figure 4.10: previous page. **Bulk rock Zn-Pb-Cu assays and Cu/Zn ratio vs. depth.** Location of mudstone intervals are shown as brown boxes. A) Zn assays, note most enrichments are concentrated in areas of pyritic mudstone. B) Pb assays, note enrichments are concentrated entirely in areas of pyritic mudstone. C) Cu assays, note most enrichments are occurring outside of pyritic mudstone. D) Cu/Zn ratio, locations of high Cu/Zn are denoted with their corresponding vein style. Note high and low Cu/Zn values are occurring within a very close proximity to each other.

Zn, Pb, Cu and Cu/Zn ratio were examined with depth to determine whether they were varying between mineralized areas (Figure 4.10). Samples in areas with high Zn either relate to the host lithology pyritic mudstone or sphalerite bearing veins which commonly show potassic alteration styles (samples 97.76, 109.16, 262.07, 284.15, 294.47, 295.55, 318.80, 336.10 and 360.75)(Figure 4.10A and D). Whereas areas of high Pb follow mudstone distribution closely (Figure 4.10B), high Cu assays correspond to samples where veins IA, IV and V were present (Figure 4.10C and D). Where type IA veins show a potassic rind (within QFP and mudstone host units) the ratios show a great range in Zn and Cu values at a cm scale.

4.4.5 Fluid Inclusion Study

4.4.1 Fluid Inclusion Petrography

Samples chosen for petrographic study were those containing calcite-hosted fluid inclusion assemblages (FIA) in vein types V. Quartz-hosted FIAs in other vein types (specifically type IA) were not workable because quartz/calcite grains were cloudy and inclusions were far too small ($<3\mu\text{m}$) to obtain accurate measurements.

There was no evidence of growth zoning in the calcite crystals or oriented entrapment of the inclusions. However, inclusions do not occur along healed fractures so primary origin is suggested. Two types of calcite host inclusions (calcite A and calcite B). There is a large variation in inclusion size and the number of phases present at room temperature within single assemblages for both calcite types. Inclusions in calcite A contain an aqueous fluid and a vapor bubble; some inclusions contain a solid (halite) phase. Most inclusions are smaller than those within calcite B, with an average diameter of $\sim 10\mu\text{m}$;

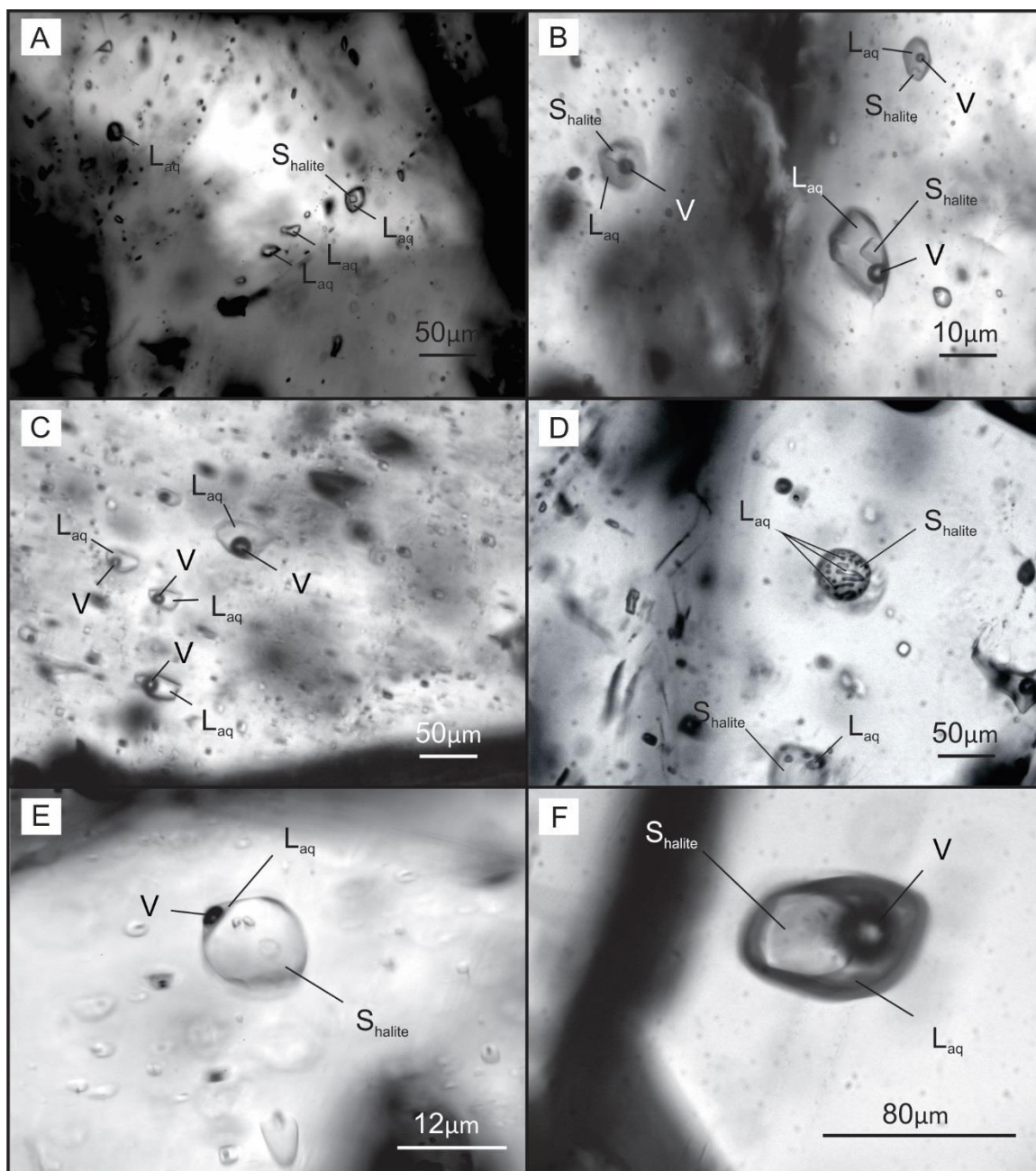


Figure 4.11: Fluid inclusion petrography and example compositional types. A) Photomicrograph of multiple one phase liquid (L_{aq}) inclusions and a single two phase $L_{aq} + S_{halite}$ in calcite B. B) Photomicrograph of a fluid inclusion assemblage (FIA) featuring three phase vapor (V) + $L_{aq} + S_{halite}$ inclusions in calcite B. C) Photomicrograph of two phase V + L_{aq} inclusions in calcite A. D) Photomicrograph of a two phase $L_{aq} + S_{halite}$ inclusion where the inclusion is super salt saturated (almost 100% halite) with liquid films running along the inclusion surface in calcite B. E) Photomicrograph of a FIA, three phase V + $L_{aq} + S_{halite}$ where the inclusion is super salt saturated (99% halite) with a very small pouch of liquid and vapor in calcite B. F) Photomicrograph of a large (80 μ m) three phase V + $L_{aq} + S_{halite}$ – salt saturated inclusion in calcite B.

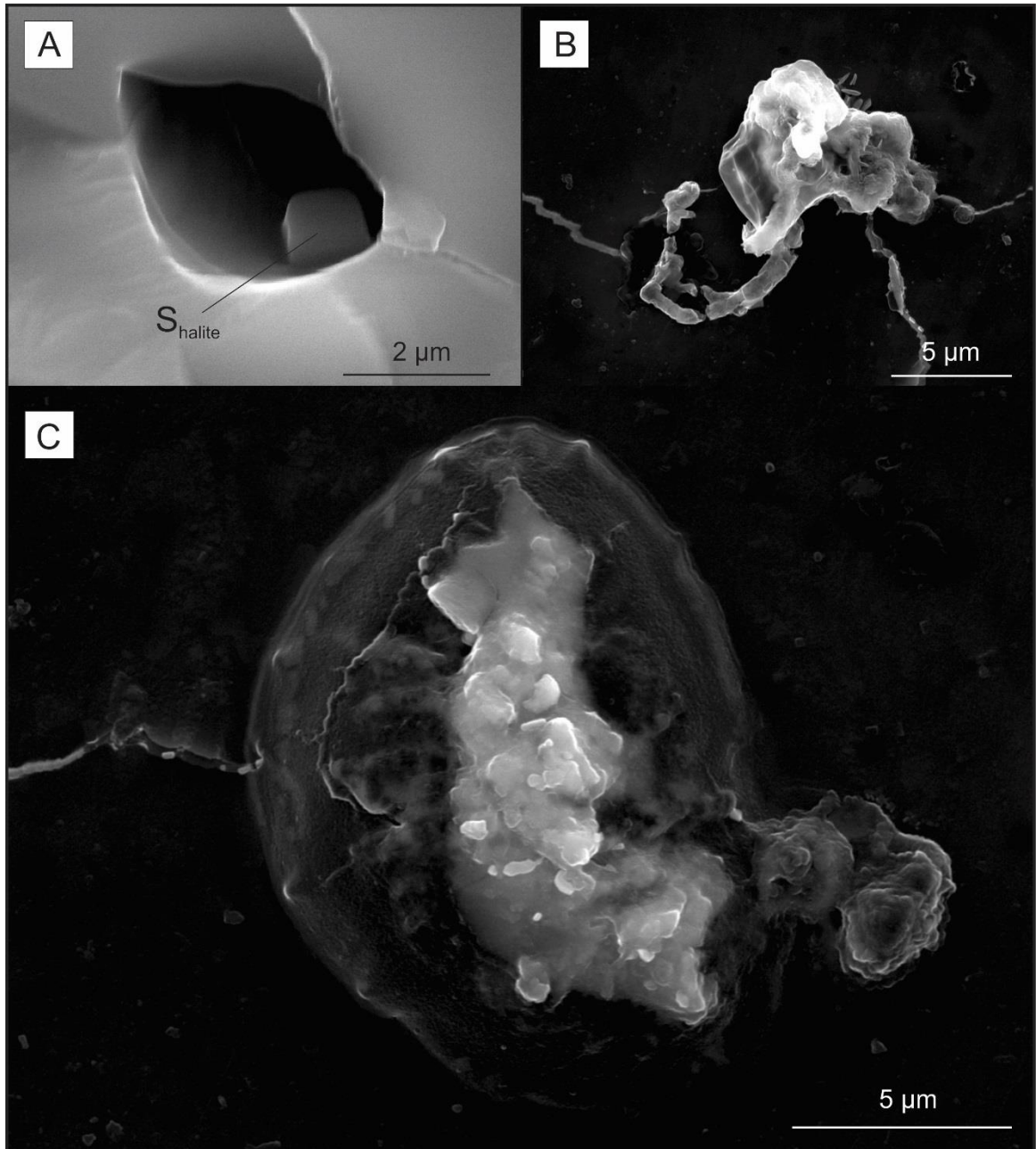


Figure 4.12: SE High-resolution images of decrepitate salt mounds on calcite crystal surfaces. A) Inclusion cavity with a defined S_{halite} crystal within. B) A Ca-rich salt mound on the sample surface. C) A large salt mound on calcite surface.

they are anhedral in shape and contain approximately 90% liquid. Calcite B was also host to phase-variant fluid inclusions. On average they have a diameter of 50 μ m, are anhedral in shape and contained ~75 - 95% liquid. Microscopic observations of fluid inclusions within both A and B calcite at room temperature (20°C) revealed four phase variations: (i) one phase, L_{aq} (Figure 4.11A); (ii) two phase, $L_{aq} + V$ (Figure 4.11C); (iii) two phase, $L_{aq} + S_{halite}$ (Figure 4.11 A and D) and (iv) three phase, $L_{aq}+V+S_{halite}$ (Figure 4.11 B, E and F).

Inclusions containing a halite crystal at room temperature have varying salt crystal volume proportions. Figure 4.11 A and B show examples of $L_{aq}+S_{halite}$ and $L_{aq}+V+S_{halite}$ inclusions showing true daughter crystals. Figure 4.11 D, E and F show examples where the inclusion fluid was salt-saturated at the time of entrapment, now containing accidental halite crystals (ranging from 50-99 vol. % halite).

In order to detect the presence of other solutes in the inclusions calcite chips were heated to >500°C in order to decrepitate the inclusions and create salt mounds on the sample surface. SE imaging was used to produce high-resolution image of the sample surface in order to locate salt mounds. Once located by SE imaging the composition of the salt mound was determined by SEM-EDS. Salt mounds had the following composition: Ca (52.1 wt. %), Cl (14.5 wt. %), Na (12.3 wt. %), Fe (8.1 wt. %), Al (4.2 wt. %), K (3.9 wt. %), Mg (2.1 wt. %), F (1.9 wt. %) and S (0.8 wt. %).

Aside from halite, identified by SEM-EDS, a rare scandium yttrium silicate $(Sc,Y)_2Si_2O_7$ (thortveitite) was identified within opened fluid inclusions (Figure 4.13). Due to the open-space filling habit of the thortveitite and its presence in not all inclusions, it was considered to be an accidentally trapped phase, coeval with surrounding fluid and acted as a nucleation point for inclusion formation.

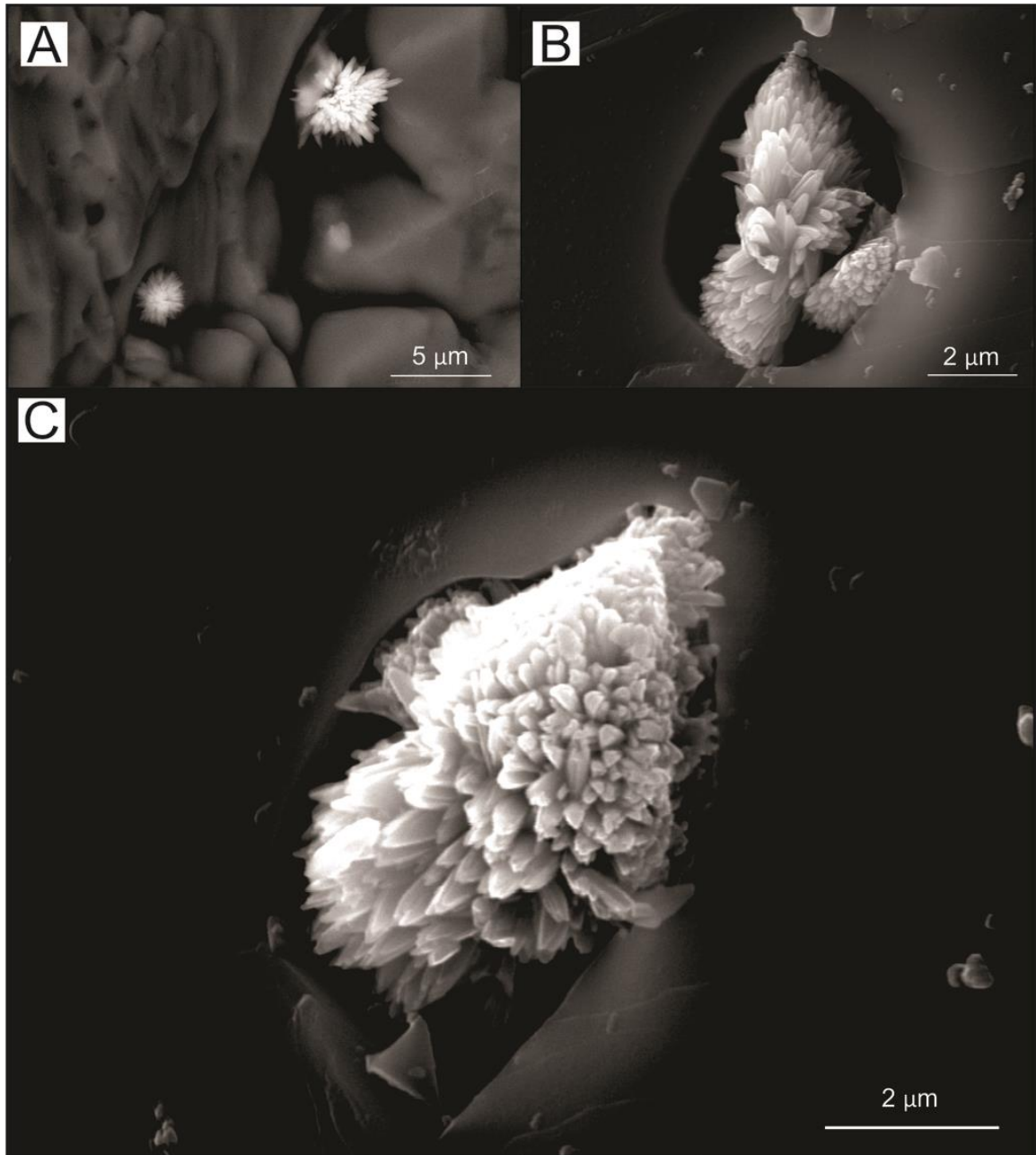


Figure 4.13: SE high-resolution images of thortveitite within fluid inclusions. A) SE high-resolution image of two thortveitite grains ranging in diameter from 2-5 μm . Both are found within fluid inclusions. B) SE high-resolution image of a thortveitite grain ($\sim 5\ \mu\text{m}$) within a fluid inclusion C) SE high-resolution images of a thortveitite grain ($\sim 5\ \mu\text{m}$) within a fluid inclusion.

Thortveitite is a relatively rare scandium-bearing mineral associated with fluoride-rich fluids and is commonly magmatic in origin (Foord et al. 1993). Worldwide, thortveitite has been identified fewer than a dozen times (Potter and Mitchell, 2005, Dunn, 1995, Foord et al. 1993, Voloshin *et al.* 1991, Bianchi *et al.* 1988, Yamada *et al.* 1980, Amli 1977, Mathiesen 1970, Oftedal 1969, Ito *et al.* 1968, Phan *et al.* 1967, Sakurai *et al.* 1962, Neumann 1961). Globally fluorite is commonly texturally associated with thortveitite but their relationship is poorly understood. Foord et al. (1993) believed that thortveitite occurrences are related to pegmatites, implying that fluid at Taylors Brook may have interacted with felsic rocks or had some contribution of (felsic) magmatic fluid added.

4.4.3 Microthermometry

Some FIAs in calcite showed ice formation at temperatures (T_F^{ice}) between -44.3 and -103.2°C but the majority showed no visible bubble contraction or ice formation so they were cooled to a temperature of -180 °C to promote freezing in the absence of visible evidence. The very low freezing temperatures suggest a high salinity fluid containing divalent cations. This is in agreement with the BSE decrepitate mound analysis that shows the presence of Ca^{2+} , Mg^{2+} and Fe^{2+} .

As the inclusions were heated there was a darkening of the inclusion associated with eutectic melting observed at temperatures between -65.7 to -26 °C with an average of -49°C (n=37). Between these temperatures most inclusions were three phase consisting of $L_{aq} + V + S_{hydrohalite}$. This eutectic range combined with hydrohalite dissolution temperatures (-52 to -12 °C, n=35) restricts the persisting solid (above 0°C) to a $CaCl_2$ -

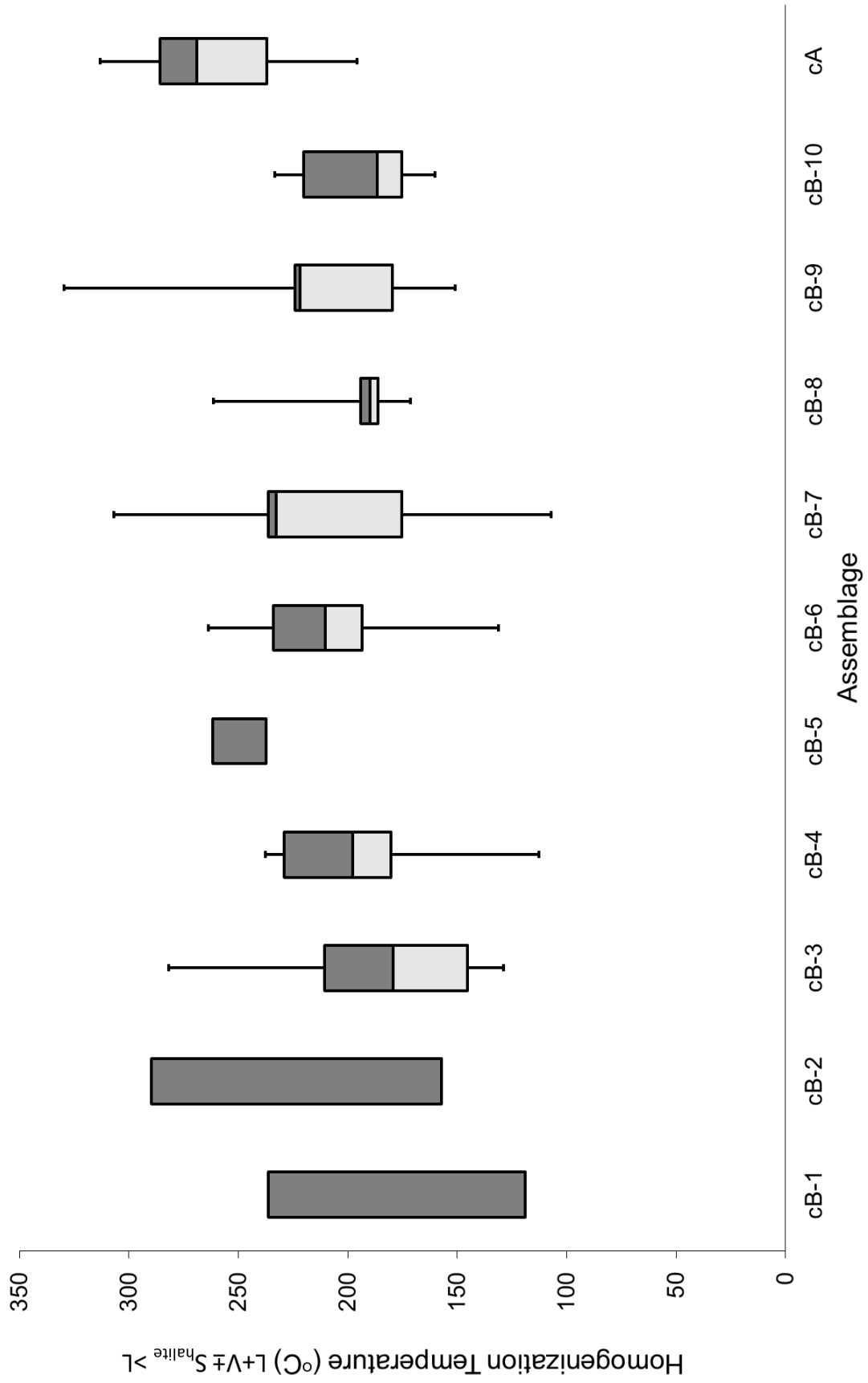


Figure 4.14: previous page. **Box and whisker plot of homogenization temperatures.** Homogenization temperatures of a vapor phase in calcite A (cA) and calcite B (cB) fluid inclusion assemblages. Upon homogenization, the two-phase (liquid and vapor) transitioned into a single-phase fluid. The upper limit of each box represents the upper quartile (Q3), or maximum, temperature of the FIA; whereas, the lower limit of each box represents the lower quartile (Q1), or minimum, temperature of the FIA. The whiskers represent maximum and minimum temperatures outliers and were not considered when describing the lower and upper T_h for each FIA.

hydrate. The inclusions contained a solid at temperatures ranging between 12 to 35°C (n=38), leaving $L_{aq}+V$ only.

Homogenization temperatures (T_h) were measured on inclusions in both calcite A and B; Calcite A had one measured FIA in which inclusions homogenized to liquid between 237 °C (lower quartile, Q1) to 285 °C (upper quartile, Q3) and an average of 266.4 °C (n=19)(Figure 4.14). Calcite B FIAs homogenized to a liquid between 117 °C (lower quartile, Q1) to 264 °C (upper quartile, Q3) with an average of 198.5 °C (n=85) (Figure 4.14). In three phase inclusions where halite crystals were present the halite dissolved at temperatures ranging 202.9 to >310 °C (n=17). As the inclusions cooled there was no halite nucleation which an indication of metastability.

Final ice melting temperatures (T_{mice}), antarcticite melting temperatures (T_{mant}), and rarely, halite dissolution temperatures (T_{mhl}) (Table 4.2) were used to calculate bulk inclusion salinities (wt% total salt, NaCl and CaCl₂) using the model of Steele-MacInnis et al. (2011). In inclusions where only CaCl₂-hydrate melting was observed, CaCl₂ wt% equivalents were estimated graphically using the UNIQUAC model for the binary CaCl₂-H₂O systems (Sander *et al.*, 1986). The programs BULK and ISOC (Bakker, 2003) were used to model bulk fluid densities and isochores for fluids in the CaCl₂-NaCl-H₂O and CaCl₂-H₂O systems. Figure 4.15 shows the resulting isochore range with metamorphic facies to understand and enforce hydrothermal reactions and the pyrrhotite-pyrite boundary temperature constraint of 400°C.

Table 4.2: Microthermometric and calculated salinity data.

Host	Hydrohalite melting (°C)	CaCl ₂ -hydrate melting (°C)	Vapor Homogenization (°C)	Halite Dissolution (°C)	CaCl ₂ equivalent (wt%)	CaCl ₂ (wt%)	NaCl (wt%)	S wt%	Density (g/cc)	Phases (20°C)
CalB		31.7		205.0		43.5	10.0	53.5		L+V+H
CalB		31.7	164.2	204.8		43.6	9.9	53.5	1.51	L+V+H
CalB	-23.7	35.4	227.2			49.8	0.1	49.8	1.67	L+V
CalB		12.2	203.9		40				1.38	L+V+H
CalB	-37.6	85.9	233.0		59				2.69	L+V
CalB		28.0	178.8		47				1.56	L+V
CalB		28.4	164.9	209.8		42.2	10.8	53.0	1.47	L+V+H
CalB		28.4	186.8	214.8		42.0	11.3	53.3	1.46	L+V+H
CalB	-12.7	28.4	200.1		47				1.54	L+V
CalB	-12.2	28.4	186.1		47				1.55	L+V
CalB	-32.5	21.2	220.0			42.2	0.7	42.9	1.37	L+V
CalB	-32.5	21.2	232.5			42.2	0.7	42.9	1.37	L+V
CalB		31.1	235.2		50				1.65	L+V
CalB		17.3	162.2		42				1.40	L+V
CalB		17.9	165.5		42				1.40	L+V
CalB		19.9	179.8		43				1.42	L+V
CalB	-52.0	22.3	223.0			42.8	0.7	43.4	1.39	V+L
CalB	-52.2	22.0	198.2			42.6	0.7	43.3	1.40	V+L
CalB	-52.0	21.5	178.9			42.2	0.7	43.0	1.40	V+L
CalB		22.9	104.9		44				1.50	V+L
CalB		15.5	111.5		41				1.41	V+L
CalB		21.6	235.7		43				1.39	V+L+H
CalB		25.2	196.6		46				1.49	V+L
CalB		18.2		274.6		32.5	20.0	52.5		V+L+H
CalB		24.5	232.1		46				1.43	
CalB		24.5	233.0		46				1.45	
CalB		24.5	304.4		46				1.39	
CalB		26.4	208.8		46				0.85	V+L+H
CalA	-12.7	76.1	209.6		58				2.52	V+L+H
CalA		28.4	295.5		47				1.49	V+L
CalA		26.4	229.0		46				1.48	V+L
CalB		25.9	221.5		46				1.47	V+L
CalB		26.1	253.0		46				1.47	V+L
Averages	-23.8	31.3	210.9	210.8	46	42.3	6.0	48.3	1.51	

note: inclusions in column CaCl₂ equivalent (wt%) were estimated using UNIQUAC model for the binary CaCl₂-H₂O systems (Sander *et al.*, 1986). Inclusions with calculated CaCl₂ and NaCl wt% were derived from the model of Steele-MacInnis *et al.* (2011).

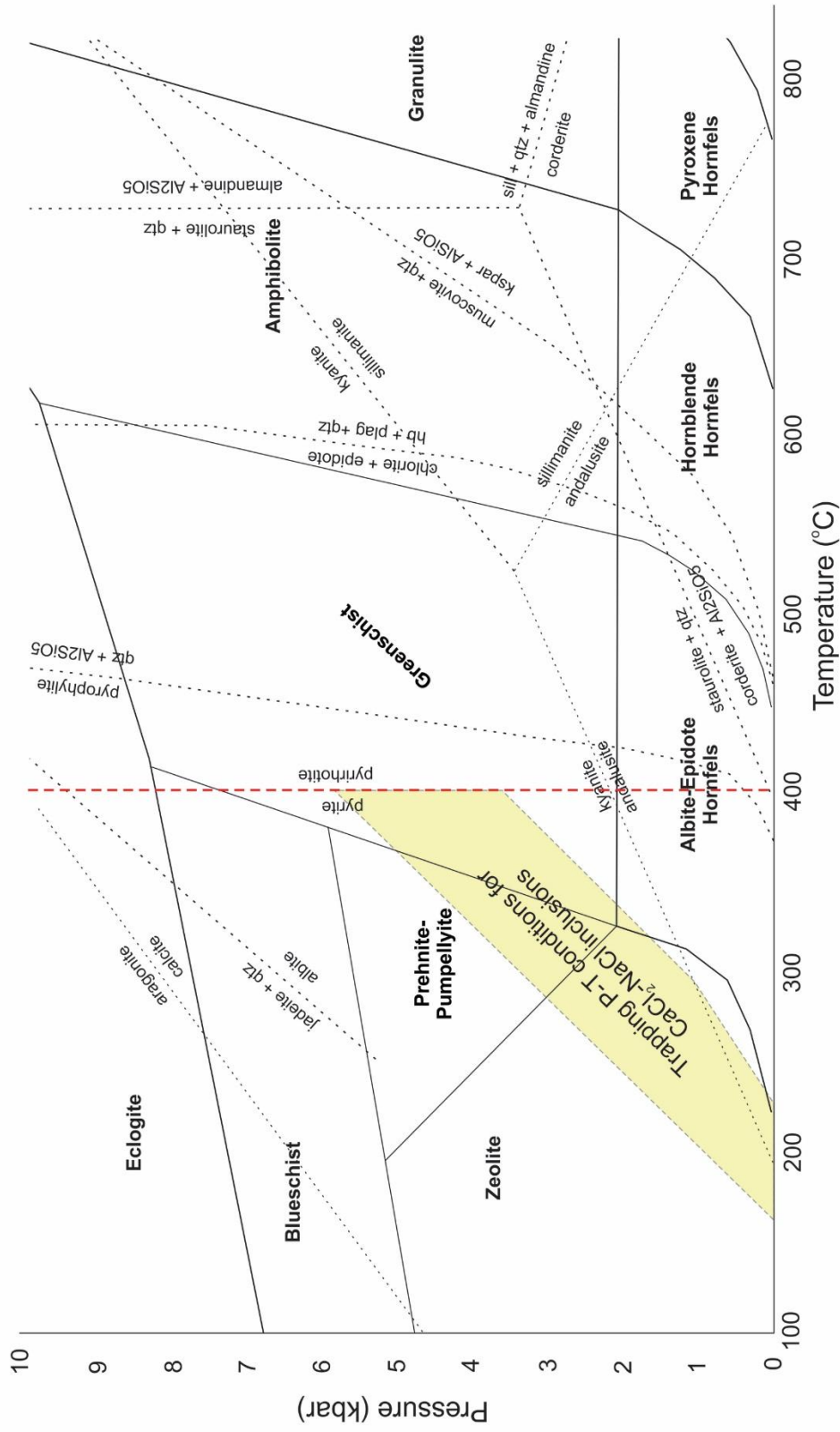


Figure 4.15: Trapping pressure-temperature constraints for CaCl₂-NaCl-H₂O inclusions. The metamorphic facies are shown to help enforce hydrothermal alteration assemblages at temperatures and pressures seen within the mineralizing vein styles. Also plotted in red is the pyrite - pyrrhotite stability boundary, obtained by analysis of pyrrhotite in the pyritic mudstones and using T-fS₂ compositional relations from Toulmin and Barton 1964. This constrains the maximum temperature of mineralization to 400°C and 5kbar.

5.0 Discussion

5.1 Mineralization and hydrothermal vein paragenesis

Based on the cross cutting relationships and bulk geochemistry data the hydrothermal system at Taylors Brook involved the circulation of a Zn-Cu-Pb-Ag-Te rich fluid leading to mineralized type IA veins. The alteration assemblages associated with this earliest mineralizing event are propylitic-potassic. Zoning in epidote in the veins is caused by variations in the Fe/Al ratio suggesting that fO_2 is fluctuating causing compositional zoning and replacement textures among hydrothermal minerals, as well as deposition of Zn-Pb-Cu sulphides. Also seen in the mineralized sulphidic mudstones is pyrrhotite altered to symthite, caused by fluctuating fS_2 or cooling, or both. Both the pyrrhotite-smythite replacement reaction and the epidote zoning is important features because they suggest that a combination of cooling, fluctuating fO_2 and fS_2 were important steps in mineralization. These parameters rendered the metals insoluble causing their precipitation within vein type IA and in the mudstones. Textural observations conclude that sphalerite, galena and chalcopyrite were precipitated late, after pyrrhotite altered to smythite.

The slightly later subset of barren type IB vein has an alteration assemblage of epidotization and silicification. The hydrothermal fluid must still be enriched with Mg-Al-Fe, a high Na content due to loss of K associated with the preceding potassic alteration, and also added secondary silica.

Type II veins have associated alteration assemblages are serpentinization- propylitic. The hydrothermal alteration resulted in forsterite and orthopyroxene reacting to produce

serpentine and talc (Wolfgang *et al.* 2006) with the fluid enriched in Mg-Al-Fe. Type III veins have a propylitic alteration style associated with a fluid enriched in Mg-Al-Fe. Type IV veins have a potassic-carbonatization-zeolitic assemblage indicating a fluid rich in Al-Si-K-Na but deprived of Fe-Mg-Ca. The presence of analcime also indicates the fluid originated from fluids in equilibrium with silica-undersaturated rocks and is likely occurring at relatively low temperatures compared to earlier vein styles.

The fluid chemistry undergoes a notable change during Type V vein deposition. The presence of thortveitite $(\text{Sc,Y})_2\text{Si}_2\text{O}_7$ indicates the fluid has some evolved felsic magmatic contribution. The fluid was enriched in Cu-Pb-Zn-Sc-Y-Ag-Te and had a bulk composition consistent with a Ca-Na-Cl hypersaline brine. The alteration assemblages associated with this late mineralizing event are silicification-carbonatization. Barren type VI record a late silicification distributed throughout the entire system.

5.2 Fluid origin in type V veins

In the interest of fluid origin and based on the results of the microthermometry and salinity studies, it was concluded that the trapped fluid in type V veins was an unusually high salinity and rich in divalent cations and anions (ie. Ca^{2+} , Mg^{2+} , S^{-2}) but particularly enriched in Ca. This distinct fluid composition can only be formed naturally under specific conditions and is very atypical of VMS fluids which are typically much lower in salinity, Na-dominant and commonly show evidence of boiling (Crowe, 1992 and Ohmoto, 1996).

There are four conceivable explanations for this Ca rich fluid. The first is boiling; if boiling of a Ca-rich fluid occurs the fluid can become supersaturated in salt as the liquid

(water) vaporizes. Eventually after some boiling, salt crystals would start to form in the system as the solution left from boiling becomes supersaturated in salt. In the case of the Taylors Brook showing this is most likely not the case because the fluid inclusions present do not show variable phase ratios consistent with boiling (Steele-MacInnis, 2012).

A second process that could lead to the formation of Ca-rich fluid is a cation exchange between Na and Ca; if a fluid initially carrying Na passes through the Ca-rich (Ca-plagioclase, Ca-pyroxene) volcanic country rocks can exchange between fluid and rock as temperature decreases. This would cause the fluid to become increasingly Ca-rich. (Hannachi, 2009) For example, if albitization (formation of Na-rich plagioclase) occurs, the Na^+ from a fluid is needed but as it goes into the structure of the feldspar, Ca^{2+} is released back into the fluid. Within the sample suite no albitization is observed suggesting this is not the case at the Taylors Brook showing.

A third explanation is evaporation of Ca rich paleoseawater; CaCl_2 basinal brines inherited their chemistry and salinity from evaporated paleoseawaters when Earth's oceans were Ca-rich and SO_4 -poor (CaCl_2 seas) (Lowenstein *et al.*, 2003). This coincided with periods of rapid seafloor spreading and high outfluxes of mid-ocean-ridge brines rich in CaCl_2 . Later, elevated sea levels were associated with conditions that favored pore-space accumulation of marine CaCl_2 brines in marginal and interior continental basin sediments (Lowenstein *et al.*, 2003). In evaporitic sequences the first saturation of a salt phase involves halite (NaCl) and as Na is removed from solution, the Ca concentration in the residual fluid increases. Eventually, the remaining fluid becomes Ca-rich and is otherwise known as a "bittern" brine. If this bittern brine gets trapped in sediments, or if it infiltrates basement rocks, fluid inclusions containing the Ca-rich brine can be trapped

(Spencer, 1987). However, a fluid this Ca-rich would be very late stage in the evaporitic sequence when residual fluid volumes are very low. The hydrothermal fluid needed for mineralizing and altering processes seen at the Taylors Brook showing greatly exceeds that generated in a late stage evaporitic sequence.

The fourth and most likely explanation is dissolution of an existing evaporitic sequence containing gypsum or carbonates by a hydrothermal fluid causing an increase in Ca fluid concentration. This is a probable explanation because bedrock surrounding Bras d'Or lake, immediately north of the Taylors Brook showing comprises the Carboniferous (330-338 Ma) upper (mudstone, sandstone, minor conglomerate, gypsum and shallow marine limestone, 100-700m), middle (halite, anhydrite, gypsum, mudstone, siltstone, sandstone and conglomerate, 300m) and lower (anhydrite, salt, gypsum, shale, marine dolostone and limestone, >150m) Windsor Group. However, this does not explain the high salinity of the fluid since dissolution of halides would contribute abundant Na and K to the evolving fluid but not Ca. Dissolution of a halide-gypsum-carbonate package could source both the observed salinity and Ca rich nature of the fluid.

5.3 Implication for mineral exploration

Exploration for this style of mineralization should focus in areas in proximity evaporates. Since these target areas help to explain the formation of the Ca-rich fluids that were responsible for metal transport.

It is also important to identify alteration styles associated with mineralization, notably potassic-propylitic-silicification-carbonatization assemblages. These alteration styles may

be identified directly during core logging. Thin section petrography is most appropriate because of the complexity of vein styles identified.

Once alteration styles/assemblages are confirmed, it is important to identify pathfinder elements associated with the hydrothermal system by XRF or SEM analysis; notably, K-Cu-Zn-Pb-Sc-Y-Ag-Te-S. The element most impacted by hydrothermal remobilization after pyritic mudstone mineralization is Cu (Figure 4.10).

5.4 Comparison to the Stirling deposit and other VMS systems

The Taylors Brook showing and Stirling deposit show similar host rock lithologies, alteration and mineralization styles. However, the most prominent alteration assemblage (quartz-talc-carbonate [QTC]) found at the Stirling deposit (O'Reily, 2008) is not present at the Taylors Brook showing. This alteration assemblage is particularly important due to its association with sulphide mineralization at the Stirling deposit. It is thought that the QTC rock is either an exhalative layer related to the same processes that gave rise to the massive sulphide lenses as hydrothermal plumes vented out onto the sea floor, or hydrothermal replacement of the host volcanics (O'Reilly, 2008). Perhaps, the Taylors Brook showing is a distal expression of these products.

The Duck Pond-Lemarchant deposits in Newfoundland are also comparable to the Taylors Brook showing. Both Duck pond and Lemarchant are situated in the northeast trending Tally Pond volcanic belt. Duck Pond is a Cu-Zn-Pb-Ag-Au VMS deposit hosted within felsic volcanic and volcanoclastic rocks and sedimentary rocks with mineralization occurring in numerous lenses (Piercey *et al.*, 2012). Textural associations are interpreted

Table 5.1: A comparison of the Taylors Brook showing to other VMS systems

Deposit	Host Rocks	Key Alteration styles	Mineralization type
Taylors Brook occurrence Cu-Zn-Pb-S (Sc, Y, Ag, Te) (This study)	basalt/andesite flows, quartz feldspar porphyry, pyrite- pyrrhotite bearing mudstone, cherty mudstone, chert, andesitic lapilli tuff and ash.	propylitic-potassic	Zn (Pb) mudstones with overprinting vein hosted Cu (Zn)
Stirling deposit Zn-Pb-Cu-Fe(Ag,Au) (Kontak, 1997; O'Reily 2008)	monolithic felsic breccia, pyritiferous-cherty zones, chert, basalt and basaltic andesite flows, felsic tuff, rhyolite porphyry, gabbro, and sedimentary rocks.	quartz-talc-carbonate (QTC)	multiple sulphide lenses (Zn-Cu-Pb)
Duck Pond deposit Cu-Zn-Pb (Ag,Au) (Piercey et al., 2012)	rhyolite flows and tuff breccias, pyrite-pyrrhotite bearing mudstone	sericite-silicification- chloritization	multiple sulphide lenses(Cu-Zn-Pb), grading inward to a Cu-rich center
Lemarchant deposit Cu-Zn-Pb (Ag,Au) (Squires et al., 1991, 2001)	felsic volcanic breccias, flows, lapillistone and lapilli tuffs, pyrite- pyrrhotite bearing mudstone	sericite-silicification- chloritization	semi-massive to massive sulphide (Cu-Zn-Pb) mineralization

to represent replacement-style mineralization (Squires et al., 1991, 2001). The Lemarchant deposit is a Cu-Pb-Zn-Ag-Au VMS deposit hosted within felsic volcanic breccias, flows, sedimentary rocks, lapillistone and lapilli tuffs with semi-massive to massive mineralization (Fraser et al., 2012). Both are classified bimodal felsic type VMS (formerly Kuroko-type). They are polymetallic in that they contain approximately equal enrichments in all three ore metals (Zn, Cu and Pb).

The Taylors Brook showing has features consistent with the bimodal felsic type VMS classification but its fluid characteristics from late mineralizing veins (type V) are inconsistent with VMS and are consistent with sediment hosted base metal deposits formed from Ca-rich basinal brines (eg. Mt. Isa, Australia: Heinrich et al., 1995)

4.0 Conclusions

1. The hydrothermal system consists of 7 vein types and their relative order of formation is type IA, IB, II, III, IV, V and VI veins.
2. Alteration styles associated with mineralization events are potassic-propylitic-silicification-carbonatization.
3. Replacement textures and compositional zoning in epidote is caused by varying Fe/Al, corresponding to varying $\text{Fe}^{3+}/\text{Fe}^{2+}$ ratios in associated fluids. This suggests that the $f\text{O}_2$ of the system was fluctuating, an important mechanism for metal precipitation.
4. The presence of thortveitite in type V veins indicates that the fluid at the Taylors Brook showing must have passed through felsic rocks or had some contribution of felsic magmatic fluid, and indicates that Sc and Y were mobile.

5. In the sulphidic mudstones, a replacement reaction from pyrrhotite to smythite occurred either by cooling or via an influx of S_2 into the system. Zn-Pb(Cu) sulphide precipitation in the mudstones occurred after smythite formation.
6. Equilibrium between pyrite and pyrrhotite only occurs within a specific fS_2 and temperature range. This constrains the maximum temperature of mineralization to 400°C for the pyrrhotite composition at Taylors Brook.
7. Bulk rock analyses identified Cu-Zn(Pb) and Zn,Pb(Cu) anomalies to be associated with type IA veins and sulphidic mudstone, respectively. Bulk rock analyses also showed Cu/Zn ratios to be varying widely with depth and from vein type to vein type reflecting Cu remobilization.
8. Fluid inclusion petrography in type V veins identified four phase variations within calcite at room temperature (20°C). Inclusions containing a halite crystal at room temperature have varying salt crystal volume proportions, with some clearly containing accidentally trapped salt crystals (ranging from 80-95 vol. % halite). This indicates that the fluid was salt saturated at the time of entrapment.
9. Microthermometry shows hydrohalite melting temperatures averaging -23.8°C, antarcticite melting temperatures averaging 31.3°C and halite dissolution temperatures averaging 210.90°C. Homogenization or minimum trapping temperature averaged 210.90°C, and was combined with fluid inclusion salinities and densities to create isochores. Isochoric data combined with pyrrhotite-pyrite equilibrium constrains a maximum pressure to 5 kbar and temperature to 400°C.

11. The most likely source for a high Ca-brine is either evaporation of a Ca-rich paleoseawater or dissolution of gypsum/carbonate/halide units, possibly the Windsor group. However, if this is true, then the mineralizing fluid must significantly post-date the associated host rocks, which is inconsistent with a VMS system.

12. The Taylors Brook showing has features consistent with the bimodal felsic type VMS classification but its fluid characteristics from late mineralizing veins (type V) are inconsistent with VMS and are consistent with sediment hosted base metal deposits formed from Ca-rich basinal brines (eg. Mt. Isa, Australia: Heinrich *et al.*, 1995)

5.0 References

- Akli, R. 1977: Carbonatites, a possible source of scandium as indicated by Sc mineralization in the Fen peralkaline complex, Southern Norway; *Economic Geology*, v.72, p.855-869
- Bach, W., Paulick, H., Garrido, C., Ildefonse, B., Meurer, W. and Humphris, S. 2006: Unraveling the sequence of serpentinization reactions: petrography, mineral chemistry, and petrophysics of serpentinites from MAR 15°N (ODP Leg 209, Site 1274); *Geophysical Research Letters*, v.33, p.1-4
- Bailes, A. and Galley, A. 1999: Evolution of the Paleoproterozoic Snow Lake arc assemblage and geodynamic setting for associated volcanic-hosted massive sulphide deposits, Flin Flon Belt, Manitoba, Canada; *Canadian Journal of Earth Sciences*, v.36, p.1789-1805
- Bakker, R. 2003: Package FLUIDS 1. Computer programs for analysis of fluid inclusion data and for modelling bulk fluid properties; *Chemical Geology*, v. 194, p.3-23
- Bianchi, R., Pilati, T., Diella, V., Gramaccioli, C.M. and Mannucci, G. 1988: A re-examination of thortveitite; *American Mineral*, v.73, p.601-607
- Bonnet, A. L., and Corriveau, L. 2007: Alteration vectors to metamorphosed hydrothermal systems in gneissic terranes, in Goodfellow, W.D., ed., *Mineral deposits of Canada—A synthesis of major deposit types, district metallogeny, the evolution of geological provinces, and exploration methods*: Geological Association of Canada, Mineral Deposits Division, Special Publication No. 5, p. 1035–1049.
- Canet, C., Franco S., Prol-Ledesma, R., González-Partida, E., and Villanueva-Estrada R. 2011: A model of boiling for fluid inclusion studies: Application to the Bolaños Ag–Au–Pb–Zn epithermal deposit, Western Mexico; *Journal of Geochemical Exploration*, v.10, i.2, p.118-125
- Carr, P. and Cathles III, L. 2008: On the Size and Spacing of Volcanogenic Massive Sulfide Deposits within a District with Application to the Matagami District, Quebec; *Society of Economic Geologists*, v.103, n.7, p.1396-1409
- Crowe, D., Nelson, S. Shanks, W and Valley, J. 1992: Geology and geochemistry of volcanogenic massive sulfide deposits and related igneous rocks, Prince William Sound, south-central Alaska; *Economic Geology*, vol.87, n.7, p.1722-1746

- DNR Mineral Resources Branch, 2010: Geology Matters 2010: Growing the Economy; Nova Scotia Minerals Update, v.24, n.4
- Dunn, P.J. 1995: Franklin and Sterling Hill New Jersey: the world's most magnificent mineral deposits; Mineralogical Magazine., v.60, i.3, p.537-538
- Fraser, D. and Giroux, G. 2012: NI 43-101 Technical report and mineral resource estimate on the Lemarchant deposit, South Tally Pond VMS deposit, Central Newfoundland, Canada; Paragon Minerals Corporation
- Freeman, A., A.J.E, Bird, D. Friðleifsson G. 2009: Hydrothermal Minerals Record CO₂ Partial Pressures in the Reykjanes Geothermal System, Iceland; American Journal of Science, v.309, n.9, p.788-833
- Hannachi, C., Hamrouni, B. Dhahbi, M. 2009: Ion exchange equilibrium between cation exchange membranes and aqueous solutions of K⁺ / Na⁺ , K⁺ / Ca²⁺ , and Na⁺ / Ca²⁺; Ionics, vol.15, n.4, p.445-451
- Hannington, M.D., Santaguida, F., Kjarsgaard, I.M., and Cathles, L.M. 2003: Regional-scale hydrothermal alteration in the Central Blake River Group, western Abitibi subprovince, Canada: implications for VMS prospectivity; Mineralium Deposita, v.38, p.393-422
- Heinrich, C., Bain, J., Mernaugh T., Wyborn, L., Andrew, A. S. and Waring C. L. 1995: Fluid and mass transfer during metabasalt alteration and copper mineralization at Mount Isa, Australia; Economic Geology, v.90, p.705–730.
- Ishihara, S. 1974: Geology of Kuroko deposits; Tokyo; Society of Mining Geologists of Japan; Mining geology special issue, n.6, p.645-666
- Ito, J. and Frondel, C. 1968: Syntheses of the scandium analogues of aegirine, spodumene, andradite, and melanotekite; American Mineralogy, v.53, p.1276-1280
- King, H. 2005: Epidote; <http://geology.com/minerals/epidote.shtml>
- Kontak, D.J. 1997: Preliminary Results of Geological and Geochemical Studies at the Stirling Deposit with Emphasis on Quartz-talc-carbonate (QTC) Rock Associated with Massive Sulphide Ore; NSDNR Report of Activities, p.49-70
- Kontak, D.J. 1997: Observations on Field Relationships, Petrology and Metallogeny of Southeast Cape Breton Island; NSDNR Report of Activities, p.27-48
- Koo, J. and Mossman, D. 1975: Evaluation of primary and secondary geological processes at the Flin Flon Cu-Zn deposit, Manitoba and Saskatchewan, Canada, using factor-vector analysis of ore geochemistry; Chemical Geology, v.16, p.1-14

- Lee, K.Y. 1970: Some Rare earth mineral deposits in mainland China; U.S Geological Survey Bull., 1312N: p.N1-N34
- Lowenstein, T.A., Hardie, L.A., Timofeeff, M.N, and Demicco, R.V. 2003: Secular variation in seawater chemistry and the origin of calcium chloride basinal brines; *Geology*, v.31, no. 10, p.857-860
- Macdonald, A.S. and Barr, S.M. 1993: Geological setting and depositional environment of the Stirling Group of Southeastern Cape Breton Island, Nova Scotia; *Atlantic Geology*, v.29, p.137-147
- Mathiesen, C.O. 1970: An occurrence of unusual minerals at Bidjovagge, northern Norway; *Norges Geology Unders*, v.266, p. 86-104.
- Miller, C.K. 1979: The Geological setting and Environment of base metal deposition of the Mindamar Mine, Stirling, Richmond County, Nova Scotia; Unpublished M.Sc. thesis, Dalhousie University, Halifax, Nova Scotia, p.10-40
- Neumann, H. 1961: The scandium content of some Norwegian minerals and the formation of thortveitite, a reconnaissance survey; *Norwegian Geology Tidsskr*, v.41, p.197-210
- Ohmoto, H. 1996: Formation of volcanogenic massive sulfide deposits: The Kuroko perspective; *Ore Geology Reviews*, vol.10, n.3, p.135-177
- O'Reily, G. 2008: Volcanogenic Massive Sulphides at the Stirling Zn-Pb-Cu Deposit; Mineral Inventory Files, Nova Scotia Minerals update, p.5
- Oftedal, I. 1969: Minor elements in thortveitite; *Narsk GeoIogy*, v.49, p.77-79.
- Petts, D., Longstaffe, F., Potter, J. Barr, S., and White, C. 2012 : Regional hydrothermal alteration and ¹⁸O-depletion of the ca. 620 Ma Huntington Mountain pluton and related rocks, Cape Breton Island, Canada; *Atlantic Geology*, v.48
- Phan, K.D., Fiossy, B. and Kerjean, M. 1967: le scandium dans les minéraux et les roches encaissantes de certaines pegmatites malagaches. *Bureaux Recherche Géologie minéralogie*, v.3, p.77-97.
- Piercey, S. and Hinchey, J. 2012: Volcanogenic Massive Sulphide (VMS) deposits of Central Mobile Belt, Newfoundland; Field Trip Guidebook- B4, GAC-MAC, p.14-16
- Potter, E.G. and Mitchell, R.H. 2005: Mineralogy of the Deadhorse Creek volcanoclastic breccia complex, northwestern Ontario, Canada; *Contributions to Mineralogy and Petrology*, v.150, 9.212-229

- Reid, M.G. 2014: Petrography and Geochemistry of Drill Core from the Taylors Brook Property in the Stirling Belt, Southeastern Cape Breton Island, Nova Scotia; Unpublished B.Sc thesis, Acadia University, Wolfville, Nova Scotia. 73pgs.
- Roscoe, W.E 1986: Report on the Stirling Property Cape Breton Island, N.S for Wilco Mining Company Limited by Roscoe Mining Service Inc.; Nova Scotia Department of Natural Resources Assessment Report, p.86-156
- Sander, B., Rasmussen, P. and Fredenslund, Aa. 1986: Calculation of Solid-Liquid Equilibria in Aqueous Solutions of Nitrate Salts Using an Extended UNIQUAC Equation; *Chemical Engineering Science*, v.41, p.1197-1202
- Sakurai, K., Nagashima, K. and Kato, A. 1962: Thortveitite from Kobe Omiya Kyoto, Japan; *Bulletin of Chemical Society of Japan* v.35, p.1776-1779
- Spencer, R. 1987: Origin of Ca-Cl brines in Devonian formations, western Canada sedimentary basin; *Applied Geochemistry*, v.2, i.4, p.373-384
- Squires, G.C., Brace, T.D. and Hussey, A.M. 2001: Newfoundland's polymetallic Duck Pond deposit: Earliest Iapetan VMS mineralization formed within a sub-seafloor, carbonate-rich alteration system. In *Geology and Mineral Deposits of the Northern Dunnage Zone, Newfoundland Appalachians. Field Trip Guide A2*,: St. John's, NL, Geological Association of Canada/Mineralogical Association of Canada, p.167-187.
- Squires, G.C., MacKenzie, A.C. and MacInnis, D. 1991: Geology and genesis of the Duck Pond volcanogenic massive sulfide deposit. In *Metallogenic Framework of Base and Precious Metal Deposits, Central and Western Newfoundland*. Geological Survey of Canada, Open File 2156, p.56-64.
- Steele-MacInnis, M., Bodnar, R.J., & Naden, J. 2011: Numerical model to determine the composition of H₂O-NaCl-CaCl₂ fluid inclusions based on microthermometric and microanalytical data. *Geochimica et Cosmochimica Acta* 75, p.21-40
- Steele-MacInnis, M., Bodnar, R.J., & Lecumberri-Sanchez, P. 2012: A numerical model to estimate trapping conditions of fluid inclusions that homogenize by halite disappearance; *Geochimica et Cosmochimica Acta*, 1, vol.92, p.14-22
- Toulmin, P., and Barton, P.B. 1964: A thermodynamic study of pyrite and pyrrhotite. *Geochimica et Cosmochimica Acta*, 28, p.641-671.
- Voloshin, A.V., Gordienko, V.V. and Pakhomovskii, Y.A. 1991: Scandium mineralization and the first find of thortveitite, Sc₂Si₂O₇, in granitic pegmatites of the Kola Peninsula; *Dokl. Akad. Nauk SSSR*, v.318, no. 4, p.972-976 (in Russ.)

- Wang, X., Hou, Z., Song, Y. and Zhang, H. 2015: Geological, fluid inclusion and isotopic studies of the Baiyangping Pb–Zn–Cu–Ag polymetallic deposit, Lanping basin, Yunnan province, China; *Journal of Asian Earth Sciences*, v.111, p.853-871
- Yamada, S., Okamoto, A., Takada, M., Fujwara, T. and Takemura, M. 1980: On the Thortveitite, fluocerite, pyroxferroite and other minerals from Isango mine, Oro, Nakagun Kyoto prefecture; *Chigaku Kenkyu*, v.31, p.205-222.

Appendix A

Relationship of host lithology, alteration style and vein mineralogy to vein type.

Appendix A: Relationship of host lithology, alteration style and vein mineralogy to vein type

#	Sample Number	Sample Start (m)	Sample End (m)	Thin Section type	Host Lithology	Vein type	Alteration type	Alteration Halo	Vein Textures	Inclusions
1	82.11	82.11	82.13	Doubly Polished		V	silicification-carbonatization			
2	92.92	92.92	92.92	Polished	Basalt	VI	silicification			
3	93.80	93.80	93.95	Doubly Polished	Basalt	III	potassic			
4	95.10	95.10	95.10	Doubly Polished	Brecciated Basalt	V	silicification-carbonatization			L + H/L + V + H/V + L
5	97.76	97.76	97.95	Polished	Basalt	IV	potassic-carbonatization-zeolitic			
6	98.80	98.80	98.99	Polished	Basalt	VI	silicification			
7	99.46	99.46	99.62	Polished	Basalt	VI	silicification			
8	99.91	99.91	99.91	Doubly Polished	Basalt	IB	epidolization-silicification			
9	106.17	106.17	106.37	Doubly Polished	Brecciated Basalt	V	silicification-carbonatization			L + H/L + V + H/ V + L
11	108.90	108.90	109.16	Polished	Basalt	IV	potassic-carbonatization-zeolitic			
12	110.20A	110.20	110.39	Polished	Basalt	III	propylitic			
12	110.20B	110.20	110.39	Doubly Polished	Basalt	IB	epidolization-silicification			
13	113.30	113.30	113.30	Doubly Polished	Basalt	IA	propylitic-potassic	Potassic rind		
14	136.70	136.70	136.80	Polished	Heterolithic mafic debris flow	III	propylitic			
15	150.02	150.02	150.02	Polished	Pyritic Mudstone	III	-			
16	152.46	152.46	152.57	Doubly Polished	Mafic Dyke	II	serpentinization-propylitic			
17	155.65A	155.65	155.85	Polished	Mafic Dyke	II	serpentinization-propylitic			
17	155.65B	155.65	155.85	Doubly Polished	Mafic Dyke	II	serpentinization-propylitic			
18	158.00	158.00	158.10	Doubly Polished	Mafic Dyke	II	serpentinization-propylitic			
19	200.20	200.20	200.20	Doubly Polished	Mafic Dyke	V	silicification-carbonatization			V + L
20	220.18	220.18	220.34	Doubly Polished	QFP	IA	propylitic-potassic	Potassic rind	Vuggy	
21	238.70	238.70	238.76	Polished	Pyritic Mudstone	-	-			
22	262.07	262.07	262.30	Polished	QFP	IB	epidolization-silicification			
23	284.15	284.15	284.30	Doubly Polished	QFP	IA	propylitic-potassic	Potassic rind		
24	294.47	294.47	294.75	Doubly Polished	QFP	IA	propylitic-potassic	Potassic rind		
25	295.55	295.55	295.79	Doubly Polished	QFP	IA	propylitic-potassic	Potassic rind	Vuggy	
26	296.00	296.00	296.37	Doubly Polished	QFP	III	propylitic			
27	305.50	305.50	305.68	Doubly Polished	Pyritic Mudstone	III	propylitic			
28	318.80	318.80	319.10	Doubly Polished	QFP	IA	propylitic-potassic			Vuggy
29	324.00	324.00	324.43	Doubly Polished	QFP	IA	propylitic-potassic			
30	336.10	336.10	336.30	Doubly Polished	QFP	IA	propylitic-potassic			
31	348.73	348.73	348.73	Polished	Mafic Dyke	II	serpentinization-propylitic			
32	356.00	356.00	356.19	Doubly Polished	QFP	III	propylitic			
33	358.90	358.90	359.00	Doubly Polished	QFP	III	propylitic			
34	360.75	360.75	360.88	Polished	QFP	IA	propylitic-potassic	Potassic rind		
35	365.53	365.53	365.53	Doubly Polished	QFP	IB	epidolization-silicification			
36	380.65	380.65	380.80	Polished	Mafic Dyke	IV	potassic-carbonatization-zeolitic			
37	381.27	381.27	381.41	Polished	Mafic Dyke	IB	epidolization-silicification		Massive Intergrowth	
38	383.30	383.30	383.45	Doubly Polished	Pyritic Mudstone	III	propylitic			
39	391.80	391.80	391.93	Polished	Pyritic Mudstone	V	silicification-carbonatization			
40	401.23	401.23	401.37	Doubly Polished	Mafic Volcanoclastic	III	propylitic			
41	406.00	406.00	406.13	Doubly Polished	Mafic Volcanoclastic	IB	epidolization-silicification		Massive intergrowth	
42	412.32A	412.32	412.50	Polished	Mafic Volcanoclastic	III	propylitic			
42	412.32B	412.32	412.50	Doubly Polished	Mafic Volcanoclastic	III	propylitic			
42	412.32C	412.32	412.50	Doubly Polished	Mafic Volcanoclastic	III	propylitic			
43	415.30	415.30	415.30	Polished	Mafic Volcanoclastic	IA	propylitic-potassic			

Appendix B

Bulk geochemical data obtained via portable XRF.

Appendix B: Bulk geochemical data obtained via portable XRF.

Date	Depth (m)	Lithology	S (ppm)	Cl (ppm)	K (ppm)	Ca (ppm)	Ti (ppm)	Cr (ppm)	Mn (ppm)	Fe (ppm)	Ni (ppm)	Cu (ppm)	Zn (ppm)	As (ppm)	Rb (ppm)	Sr (ppm)	Y (ppm)	Zr (ppm)	Nb (ppm)	Ag (ppm)	Ba (ppm)	Pb (ppm)	Cu/Zn	
7/7/2015	-6.57	granite			19560	10018	1733	25	284	10516	18		21.5		59.5	779		65	3.8	46	665	7.8		
7/7/2015	-8.00	granite			19025	11314	1938	38	369	14743	17		32	8.9	71.3	758		111	6.5	44	681	4.5		
7/7/2015	-11.10	granite			16540	11443	1043	21	158	6364			15.5		51.2	901	4.3	87	2.6	24	524	10.5		
7/7/2015	-13.11	granite			32764	8803	13778	13	288	13778	13		20.6	6.1	111.5	525	12.4	117	10	54	1291	6.8		
7/7/2015	-16.40	granite			21431	12464	2539	38	390	16819	20	10	27	10.5	75.8	580	5.4	61.9	5.3	41	833	8.3	0.37	
7/7/2015	-19.45	granite			16019	9395	2547	110	471	17658	32	11	50	5.9	71.2	330	11.6	137	9.4	39	623		0.22	
7/7/2015		granite	246		30226	6085	1371	19	221	7922	17		24.1	3.6	88.6	356	4	90.1	5.5	48	1052	8		
7/7/2015	-24.15	granite			258	27255	6457	1874	22	269	10492	11	10	17.8	6	88.6	356	4	90.1	5.5	48	972		0.56
7/7/2015	-27.00	granite			242	28546	4115	1539	37	313	16781	25		32	7.4	77.7	274	5.4	66.2	5	43	775		
7/7/2015		granite			812	38273	3700	1352	31	219	8859	14		23.1	5.6	105	401	71.7	3.9	49	1061	7.7		
7/7/2015	-33.02	granite			261	44257	6312	1691	32	239	10920	16		23.5	11.6	104.4	610	5	73	5.1	52	1178	14.7	
7/7/2015	-36.00	granite			272	26266	6501	1768	25	290	12090	19		26.9	8.4	83	445	5.8	76.7	6.3	45	1114	5	
7/7/2015	-39.02	granite			491	24339	8929	1790	19	288	10660	20		22.2	6.5	69.4	368	5.7	90.1	5.1	33	690		
7/7/2015	-42.00	granite			38397	14959	1808	44	302	11629	14		40	11.7	121.6	1020		87	7.1	61	1317	17.1		
7/7/2015	-45.15	granite			64399	15598	1412	29	200	7000	15		13.1	5.4	142	377		36.2	5.1	49	1512	10.4		
7/7/2015	-48.00	granite	294		16554	7180	2702	26	313	13610	13		26.4	6.3	56.6	244		89	8.2	43	544			
7/7/2015	-51.12	granite			14632	7452	2475	21	384	15470	17		28.2	8.8	54	328	4.1	72.3	5.4	34	575			
7/7/2015	-53.92	granite			35111	13113	1514	60	201	6114			11.1	10.4	83.4	485		64.3	3.5	48	1312			
7/7/2015	-57.05	granite			24737	8649	2742	37	425	17388	23		35	13.2	78.1	456	9.4	83.1	8.1	49	1157			
7/7/2015	-60.14	fault zone	1000		11359	10011	7083	314	1194	64282	45		172	16.3	59.8	126	15.3	88.7	6.4	59	292			
7/7/2015	-65.12	fault zone	549		12529	34282	4443	350	2287	97428	33		259	22.1	27.9	91	5.9	21.8	2.7	77	297			
7/7/2015	-66.00	fault zone	789		592	5875	17030	3890	307	1876	89227	63		194	22	15.4	553	13.9	25.3	4.4	81			
7/7/2015	-69.50	fault zone	4984		627	556	164955	6254	913	61888			37	33	12.7	2653	21.2	56	13.4	74				
7/7/2015	-72.00	basalt			4037	23843	3572	139	1477	66938	23		84	10.8	11	90	8	22.5	2.6	61	46			
7/7/2015	-73.16	basalt			408	4929	33001	147	1502	64034	22		129	14.1	13.1	142	10.9	31.7	2.6	77	172			
7/7/2015	-74.00	basalt			6579	93214	3521	129	1347	54702	32		90	16.6	17.5	119	11.2	28.8	2.5	62	143			
7/7/2015	-75.00	basalt			16873	25289	2794	163	1461	54907	29		95	17.5	68.5	456	6.5	13.2	70	762				
7/7/2015	-76.10	basalt			369	8901	19995	4261	24	1148	52439	89		89	24.3	25	412	23.6	94	7.8	65	536		
7/7/2015	-77.00	basalt			401	4020	27522	2533	122	1913	61701	23		204	13	6.5	127	8.5	14.4	50	53			
7/7/2015	-78.02	basalt			24174	44028	2733	197	1854	64979	42		100	21.1	62.7	255	7.9	11.6	80	1040				
7/7/2015	-79.98	basalt			6023	101131	1638	70	1370	52885	23	16	66	24.4	10.5	416	7.8	7.9	2.4	73	150		0.24	
7/7/2015	-80.91	basalt	2068		328	16650	81136	143	1201	35377	42		319	72	11.2	46.1	267	14.3	19.9	61	747		4.4	
7/7/2015	-82.00	basalt			384	479	64919	924	63	950	45251			212	48	9.6	127	12.7	6.5	59			4.4	
7/7/2015	-82.80	basalt			213	638	1592	2167	556	10949			105	42	6.3	92.2	12.9	90.8	2.3	31	10		2.5	
7/7/2015	-85.05	Intermediate porphyry	1587		646	196584	4213	52	698	58648					21	12.2	2002	13.4	72	11	89		8	
7/7/2015	-86.87	Intermediate porphyry	2911		11634	50613	5516	71	954	52281	31	41	51	19.6	50.5	657	18.8	78	7.7	67	352		0.80	
7/7/2015	-87.75	Intermediate porphyry			7823	138719	1330	29	1181	38056	15		62	11.8	20.6	31.5	5.3	5.7	6.3	63				
7/7/2015	-88.23	basalt	401		9678	38393	2235	83	1530	47863	16		70	16.4	24	295	4.9	10.6	2.4	57	499		8.4	
7/7/2015	-89.10	basalt	21010		981	7466	47444	617	18	617	31679	560		67	35.5	17.6	341	10.2	37.6	7.3	54	220		
7/7/2015	-90.57	basalt			728	8271	24094	57	1632	71101	232		140	13.9	22.4	352	3.8	11.6	11.1	2.7	79	234		
7/7/2015	-91.10	basalt			851	5023	58041	42	2639	70373			133	16.2	13.1	426								
7/7/2015	-92.70	basalt			2223	55892	1988	56	2729	215923			182	36		237	8.7	10.9	12.1	54				
7/7/2015	-92.92	basalt	117098																					

Date	Depth (m)	Lithology	S (ppm)	Cl (ppm)	K (ppm)	Ca (ppm)	Ti (ppm)	Cr (ppm)	Mn (ppm)	Fe (ppm)	Ni (ppm)	Cu (ppm)	Zn (ppm)	As (ppm)	Rb (ppm)	Sr (ppm)	Y (ppm)	Zr (ppm)	Nb (ppm)	Ag (ppm)	Ba (ppm)	Pb (ppm)	Cu/Zn
7/7/2015	-94.88	basalt		492	5180	42495	2650	46	2057	61331		71	223	15	16.4	464	6.3	13.2	64	64	331		0.32
7/7/2015	-96.10	basalt		437	3522	44633	4739	304	1589	67732	41	39	97	18.4	10.1	259	25.2	78.5	5.1	66	84		0.40
7/7/2015	-96.86	basalt		415	3675	40131	579	11	785	23095	18	38	111	15.4	9.1	391	8	31.3	8	39	198		0.34
7/7/2015	-99.91	basalt		556	2828	29767	2334	47	1688	76334	31	88	66	19.4	4	431	5.6	13.8	2.7	69	57		
7/7/2015	-100.91	basalt		5293	66941	68997	2481	70	1454	73731	17	27	84	11.9	5.4	378	6.1	12.1	2.5	63	111		0.32
7/7/2015	-102.02	basalt		1086	2808	49226	2297	54	1155	63387	20	27	84	11.9	5.4	378	6.1	12.1	2.5	63	111		0.32
7/7/2015	-102.74	basalt	209351	1133	2029	27424	900	37	1513	349283	111	2787	119	35	17.3	31.5	201	6.5	12	122	97	38	23.42
7/7/2015	-103.16	basalt			13156	72169	2190	50	1517	71713	18	33	95	17.3	31.5	201	6.5	12	122	97	38		0.35
7/7/2015	-104.24	basalt			24770	171377	709	21	527	23532	11	11	27	8.1	73.4	183	7.7	3.9	7.7	66	820	0.41	0.41
7/7/2015	-105.64	basalt		1410	12776	9398	4755	654	1083	53109	109	205	108	29	17.3	54.5	149	27.3	53.5	61	658		0.41
7/7/2015	-106.69	basalt	244973	933	1407	12512	1137	32	1127	287603	109	205	108	29	17.3	54.5	149	27.3	53.5	61	658		0.41
7/7/2015	-107.80	basalt		577	3991	65615	2309	53	1484	41637	27	77	12.4	10.9	21.4	31.7	8.5	5.9	11.6	155	2.41		2.41
7/7/2015	-108.40	basalt		577	2329	41853	2507	55	2474	57704	18	112	157	13.7	4.1	603	6.6	14.3	69	56		0.71	0.71
7/7/2015	-109.16	basalt		396	1158	11364	3121	67	1658	71200	30	33	108	29	17.3	54.5	149	27.3	53.5	61	658		0.41
7/7/2015	-109.54	basalt		1175	1357	12274	785	26	1172	230845	109	205	108	29	17.3	54.5	149	27.3	53.5	61	658		0.41
7/7/2015	-111.10	basalt		739	7485	79530	2246	44	2492	63067	17	32	275	19	21.4	62.2	5.5	8.4	71	408			0.12
7/7/2015	-112.52	basalt	66306	518	25486	56534	2171	46	1875	90857	16	1415	115	29.5	64.9	62.2	5.5	8.4	71	408			0.12
7/7/2015	-112.76	basalt		518	2602	48921	2680	60	3985	57177	16	725	382	16.6	5.6	263	3.7	12.9	64	329			12.30
7/7/2015	-113.22	basalt	59206	1224	13534	54196	1913	71	2491	41273	16	725	382	16.6	5.6	263	3.7	12.9	64	329			12.30
7/7/2015	-113.30	basalt		2199	1557	79015	186	42	840	539460		11421	697	14	21.8	165	12	13.6	6.5	19.2	54		0.01
7/7/2015	-115.02	basalt	60419	387	1203	44023	2507	49	2808	69152	91	105	105	105	14.7	268	4.8	13.6	63	87			16.39
7/7/2015	-117.67	basalt	20496	87	5497	180767	1034	15	1619	30857	19	116	160	21.9	8.2	73	8.2	7	68				0.87
7/7/2015	-118.00	basalt	1154	582	765	6218	2603	64	795	32669	19	36	71	11.2	4.1	320	4.1	13.3	30				0.73
7/7/2015	-119.10	basalt		552	2579	59047	2464	51	1880	60581	30	36	158	12.7	3.8	261	7.5	14.5	71	99			0.51
7/7/2015	-120.30	basalt		552	1915	48594	2774	48	2065	77307	30	36	158	12.7	3.8	261	7.5	14.5	71	99			0.12
7/7/2015	-121.10	basalt		446	5122	16957	2980	66	2150	85890	18	12	34	16.1	12	403	5	15.4	64	466			0.12
7/7/2015	-122.12	basalt		400	441	224793	876	17	1122	28231	16	102	235	14.9	12.6	306	8.1	17.3	76	158			0.35
7/7/2015	-123.22	basalt	6731	489	4381	52434	2521	41	2429	53873	19	102	235	14.9	12.6	306	8.1	17.3	76	158			0.43
7/7/2015	-124.15	basalt		489	7820	69126	2269	42	1969	56905	19	25	201	17.8	52.8	63.1	6.4	10.6	89	1081			0.12
7/7/2015	-125.09	basalt		497	5898	39441	2554	57	2706	75911	23	25	201	17.8	52.8	63.1	6.4	10.6	89	1081			0.12
7/7/2015	-126.20	basalt		503	5084	61670	3221	34	1945	56149		61	78	16	10.7	242	7.7	20.2	109				1.75
7/7/2015	-127.00	basalt		306	6888	55616	2560	22	1505	35024	15	74	29	14.4	4.2	327	23.3	93.9	66	296			0.78
7/7/2015	-134.10	rhyolite		1711	29556	1457	1457	490	15796			74	29	14.4	4.2	327	23.3	93.9	66	296			0.78
7/7/2015	-135.26	heterolithic		1383	199359	2034	24	1662	42751			15	65	9.4	2.6	326	6	19.6	3.3	59	122		0.23
7/7/2015	-136.00	heterolithic		396	2549	65283	2635	30	1727	45264		15	65	9.4	2.6	326	6	19.6	3.3	59	122		0.23
7/7/2015	-136.82	rhyolite	2009	235	2247	15238	1234	8	530	144997		57	18.2	7	6.2	331	26.8	95.9	3	40	59		3.13
7/7/2015	-138.00	heterolithic		377	1217	36843	2120	27	1190	41624		2519	97	11.2	39.4	1249	23	67	100	552			23.11
7/7/2015	-139.09	heterolithic	96451	893	12792	91561	1902	33	1976	132951		83	145	14.1	29.4	169	9.8	21.2	70	379			0.57
7/7/2015	-140.21	heterolithic		686	5689	76253	3185	56	1461	59449		83	145	14.1	29.4	169	9.8	21.2	70	379			0.57
7/7/2015	-141.25	heterolithic		686	7648	35881	4001	35	1481	70045		83	145	14.1	29.4	169	9.8	21.2	70	379			0.57
7/7/2015	-142.16	heterolithic	8422	619	11494	33927	2742	21	1456	62470		83	145	14.1	29.4	169	9.8	21.2	70	379			0.57
7/7/2015	-143.10	heterolithic		7870	33595	3360	43	1729	67927			83	145	14.1	29.4	169	9.8	21.2	70	379			0.57
7/7/2015	-144.02	rhyolite		249	4595	7307	1522	37	395	14700		83	145	14.1	29.4	169	9.8	21.2	70	379			0.57
7/7/2015	-145.00	heterolithic		374	6133	18836	3404	23	1843	65486		93	99	32	5.2	194	15.4	32.1	73	80			0.94
7/7/2015	-146.00	heterolithic	28865	528	3346	58318	3289	19	2086	70952		93	99	32	5.2	194	15.4	32.1	73	80			0.94
7/7/2015	-147.10	pyrite		1202	4198	5055	1081	40	987	109903	51	12	30	9.7	9.8	156	13.5	93.8	3.5	38	171		0.40
7/7/2015	-147.50	pyrite	51981	1202	4198	5055	1081	40	987	109903	51	12	30	9.7	9.8	156	13.5	93.8	3.5	38	171		0.40
7/7/2015	-147.60	pyrite	58230	1202	4198	5055	1081	40	987	109903	51	12	30	9.7	9.8	156	13.5	93.8	3.5	38	171		0.40
7/7/2015	-147.75	pyrite	61311	592	28212	34250	2268	37	1177	70899		31	177	31	3.8	144	23.2	158	4.9	65	173	22	0.12
7/7/2015	-148.20	pyrite	27061	478	13506	24617	2260	39	1010	73547		75	167	17.8	29.7	30	118	12.1	93	144	373	0.45	0.45
7/8/2015	-149.07	pyrite	19674	1534	4878	28504	4507	56	1317	65660	15	14	511	40	9.2	156	16	77.3	9	79	277	3572	0.03

Date	Depth (m)	Lithology	S (ppm)	Cl (ppm)	K (ppm)	Ca (ppm)	Ti (ppm)	Cr (ppm)	Mn (ppm)	Fe (ppm)	Ni (ppm)	Cu (ppm)	Zn (ppm)	As (ppm)	Sr (ppm)	Y (ppm)	Zr (ppm)	Nb (ppm)	Ag (ppm)	Ba (ppm)	Pb (ppm)	Cu/Zn	
7/82015	-150.02	pyritic	17289	804	2588	15725	3392	33	350	20683		409	6906	3.4	138	64	258	8.9	60	432	381	0.06	
7/82015	-150.45	pyritic mudstone	138299	945	3364	32394	1172		1136	201141		2103	302	68	110	33	95	4.4	111	107	26	6.96	
7/82015	-151	pyritic	34282	835	4065	17283	1977	22	681	57260		123	1132	8.1	150	34	123	10.4	69	105	1170	0.11	
7/82015	-151.55	mafic debris	1223	1207	5348	1784	3288	55	1142	48932	25												
7/82015	-151.86	mafic debris	1525	487	12130	46831	3703		1567	55145		103	69	13.6	31.1	220	9.5	18.7	62	442		1.49	
7/82015	-153.1	mafic debris	1237	1237	8947	65599	4276	25	2058	58297	18												
7/82015	-153.86	mafic debris	2857	5000	7982	65999	4276	25	2058	58297	18												
7/82015	-154.7	mafic debris		816	2635	30849	3966	24	795	53606	17												
7/82015	-155.8	mafic debris	62179	607	1782	126361	2773	36	1864	91760		674	177	11.5	12.8	3.5	167	14.8	23.9	86	336	3.81	
7/82015	-157.35	mafic debris		523	36842	34632	3383	61	1793	66678	14												
7/82015	-158	mafic debris	6700	2516	853585				329	5490	54												
7/82015	-159	cherty ash			15174	942	2423		745	16516	18												
7/82015	-160.2	cherty ash	28386		22553	9437	3052	33	687	19627	11												
7/82015	-162.29	cherty ash	8846		16364	13592	2127	57	801	20366	35												
7/82015	-163.1	cherty ash		266	3693	2655	1765	16	1137	24875	28												
7/82015	-164.7	cherty ash	27068		9128	7319	1869		448	22531	12												
7/82015	-165.7	cherty ash		739	3124	17772	7698	112	2340	62793	32												
7/82015	-166.5	mafic dyke		485	11204	33674	6350	170	1992	100990	78												
7/82015	-166.95	cherty ash	10153		1965	6171	1863	27	401	18704	15												
7/82015	-168.9	cherty ash	2912		7877	8571	1625	16	810	13621	18												
7/82015	-170.26	cherty ash	1686		3091	9394	1404	8	543	12030	15												
7/82015	-171.17	cherty ash			9124	17276	2747	31	945	25132	21												
7/82015	-172	cherty ash	1183		38836	22432	2813	64	1485	16231													
7/82015	-173.5	basalt		450	28950	14990	5564	43	3612	76470	83												
7/82015	-174.2	cherty ash	1129	263	23597	12432	2095	33	1166	24167													
7/82015	-175.14	basalt			1921	62015	4862	40	2330	50533	19												
7/82015	-176.13	basalt			524	183454	2909	22	978	62414													
7/82015	-177.9	basalt			462	193834	2640		1060	70454													
7/82015	-178.36	cherty ash	1853	758	12332	9115	6316	54	1285	77045	19												
7/82015	-179.13	mafic dyke	5719		4731	34850	7664	30	2716	67787													
7/82015	-179.74	mafic dyke	2543		10659	56774	10210	40	1892	89059													
7/82015	-180	mafic dyke		395	4630	50802	7277	34	2172	61485													
7/82015	-181.01	mafic dyke		421	4133	52404	6734	27	1768	52977													
7/82015	-183	grey chert	59469		3496	5661	9946	61	1573	68642	36												
7/82015	-183.95	grey chert			3891	2097	1926	15	817	23625	14												
7/82015	-187.1	mafic dyke	16689		3960	58227	2635	323	2396	58929	35												
7/82015	-188.24	mafic dyke			1575	47755	2434	453	1682	65562	69												
7/82015	-190	mafic dyke	2981	732	5895	8430	3792	575	835	65815	118												
7/82015	-190.46	mafic dyke		585	6221	14284	3453	397	806	52008	115												
7/82015	-191.6	mafic dyke		704	4477	6781	5674	481	1040	99371	150												
7/82015	-192.3	basalt			4265	78328	2483	472	2038	72739	71												
7/82015	-193.28	basalt			14885	142263	992	172	1230	35373	33												
7/82015	-194.34	basalt		658	25110	83992	2893	540	2303	79290	48												
7/82015	-195.4	grey chert	938		4243	6584	16666	16	505	11957	15												
7/82015	-196	grey chert	1381	995	2608	14005	2119	27	513	13874	12												
7/82015	-196.95	basalt			1565	38678	2454	509	2237	60678	57												
7/82015	-198	basalt		370	2530	60590	2021	395	2027	68821	64												
7/82015	-199.23	basalt		624	1976	85467	2538	481	3873	73593	53												
7/82015	-200	basalt	19427	354	10768	15166	3343	56	638	57393	26												
7/82015	-201.1	basalt	2031	413	3825	46113	2010	382	1906	46183	26												
7/82015	-202.3	basalt			575	50123	6134	154	154	67881	76												
7/82015	-203	basalt		1243	2801	9651	7825	211	1068	79870	84												
7/82015	-203.75	cherty ash	29337		1677	9835	2357	516	332	30704	16												
7/82015	-204.34	cherty ash	49369	316	880	10949	1901	930	332	30704	16												
7/82015	-205.05	cherty ash	790		6075	6580	2196	22	529	12879	18												
7/82015	-206.6	cherty ash	6954		1823	11628	2104	23	674	12787	38												
7/82015	-207	QEP	17533		29646	11271	2819	17	660	16242	16												

Date	Depth (m)	Lithology	S (ppm)	Cl (ppm)	K (ppm)	Ca (ppm)	Ti (ppm)	Cr (ppm)	Mn (ppm)	Fe (ppm)	Ni (ppm)	Cu (ppm)	Zn (ppm)	As (ppm)	Rb (ppm)	Sr (ppm)	Y (ppm)	Zr (ppm)	Nb (ppm)	Ag (ppm)	Ba (ppm)	Pb (ppm)	Cu/Zn	
7/8/2015	-208.12	QFP	956		16156	17978	2761	19	1054	22117			20	58	8.9	41.6	339	25.5	2.9	56	592			
7/8/2015	-208.12	QFP	10416		8604	42336	2056	15	995	20077			109	51	6.2	26	236	21.7	2.9	50	549	6	0.39	
7/8/2015	-210	QFP	2656		14147	14983	1181	31	602	10304				571	3.9	43.3	136	12.8		49	1222	160	0.19	
7/8/2015	-210.33	QFP			5593	18927	1717	13	841	22170				50	3.9	17.0	195	23.9		39	464			
7/8/2015	-211.22	QFP	9885		30540	14387	2947	16	697	23223				38	7.2	47.4	203	21.5		37	1896			
7/8/2015	-212.22	QFP			3540	11357	1885		893	24873				64	6.7	10.4	378	15.3		3.5	36	376		
7/8/2015	-213.18	QFP	9260		17757	16850	2333		693	24873				47	7	42.2	241	19.6		48	713			
7/8/2015	-213.89	QFP			5000	13909	2537		775	25238				60	7.6	15.8	200	24.1		36	261			
7/8/2015	-215.07	QFP	50006		8941	49081	819	31	587	41232			11	27	12.9	19	328	29.3		69	312		0.41	
7/8/2015	-216.12	QFP	33179	354	13300	42878	1050	18	464	24562			9	55	9.3	66.1	148	25.1		78	1656			
7/8/2015	-217	QFP	6716		33182	15431	2425		859	28960				55	5.8	27.2	198	22		31	57	1049		
7/8/2015	-217.95	QFP			9420	14672	2065	10	575	19556				51	6.3	23.4	293	23		4	42	256		
7/8/2015	-219.07	QFP			6918	17027	1567	13	608	19889				6.6	5.2	70.7	224	65		56	903			
7/8/2015	-220.16	QFP			20081	41749	1251	30	1620	12803				72	6.3	19.5	314	14.5		35	328			
7/8/2015	-221	QFP			5943	7496	1796		662	20254				62	6.8	19.4	206	16.2		3.3	39	283		
7/8/2015	-222.04	QFP	11947	319	4686	17470	2451		974	21286	11		63	62	6.8	19.4	206	16.2		3.3	39	283	1.02	
7/8/2015	-223.07	mafic-dike	4883		3195	37312	1129		396	11918				16.7	4.8	7	60.8	20.9		2.5	42	74		
7/8/2015	-224	mafic-dike			3918	85457	8368	60	3408	67201	27			111	15.6	5.5	321	30.1		10.1	80	255		
7/8/2015	-224.5	mafic-dike	6071	666	8034	77685	8516	96	3276	71467	24		12	152	14.7	17.5	400	29.7		10.2	79	276	0.08	
7/8/2015	-225.3	QFP	2301	225	6923	10398	2003	36	531	12281				22.1	4.4	19.9	348	16		2.5	28	365		
7/8/2015	-226	mafic-dike	14446		5004	25306	3538	34	926	30784			17	82	7.9	12.8	309	34.4		2.5	52	217	0.21	
7/8/2015	-226.6	QFP	83993	683	7319	30263	2537	87	1163	65990			36	46	12	14.3	235	24.9		64.2	85	361	42	0.78
7/8/2015	-227.2	QFP	247079	1148	15751	28687	3099		2166	242572			1622	3860	21	66	60	43		103	318	40	0.42	
7/8/2015	-228.02	QFP	19068	288	14153	12984	2748		913	15261				148	40.6	91.9	19.2	72.9		3.2	55	389	29.4	
7/8/2015	-228.4	QFP	638804	2257	12589	30022	676		946	553368			40961	10718	30	77	46			140	115	1122	3.82	
7/8/2015	-229.32	QFP	19199		34539	27125	2472	23	923	16617				2316	11.8	88.4	193	25.8		80	2228	30	0.01	
7/8/2015	-229.92	QFP	31920		19707	34089	3652	19	1996	43188			83	310	10.1	54.2	223	29.4		3.3	50	643	11	0.27
7/8/2015	-230.9	QFP	3428		7067	12397	1243	11	759	11115				563	22.7	153	13.7	75.4		2.9	39	359	18.7	
7/8/2015	-232	QFP	4370		11027	18003	2108	11	761	13886			27	1349	27.6	169	22.1	84.4		3.6	43	479	272	0.02
7/8/2015	-233	QFP	22402	283	8820	18775	2668	11	1269	22392				96	5.3	28.3	123	25.3		3.6	47	452		
7/8/2015	-233.89	QFP			9933	44057	1791		700	12909			9	435	5.7	23.7	27.1	19.5		58	393		0.02	
7/8/2015	-241	mafic-dike	36519	612	15663	76859	3600	991	2149	73527	288		19	173	18.7	60.2	73.4	18.6		2.7	59	1723	0.11	
7/8/2015	-234.8	QFP			21740	12225	33566	22	1488	34583				141	141	29.8	136	14		66.8	33	401		
7/8/2015	-236.06	QFP	11773	536	4048	68734	5504	144	2822	60679	68		32	107	15	7.9	511	20.8		4.6	73	200	0.30	
7/8/2015	-242.25	mafic-dike			10554	16799	2243		1051	12144				156	18.1	5.5	368	27.3		78	4.5	62	121	0.42
7/8/2015	-243	mafic-dike	29463	899	2924	56692	5708	96	2589	97283	38		66	18.1	7.5	238	37.3	90.7		6.9	37	135		
7/8/2015	-243.2	pyrite			5780	28785	976	22	566	9744				18.1	4.6	66.6	96.6	11.7		6.5	61	77	2.14	
7/8/2015	-243.49	pyrite			27128	75385	5032	162	1999	49492	32		12	82	12.3	64.6	528	19.7		38	86	1507		
7/8/2015	-239.56	pyrite	5517	945	8129	5949	1999	14	430	11684	21		21	68	35.1	102	22.2	134		8.8	21	406	29.6	0.31
7/8/2015	-241	mafic-dike			1231	6096	1496	11	701	13685	22		54	75	4.4	104.7	16.3	103.5		6.7	38	113	0.72	
7/8/2015	-245.14	pyrite	4393		1231	6096	1496	11	701	13685	22		54	75	4.4	104.7	16.3	103.5		6.7	38	113	0.72	
7/8/2015	-246.06	pyrite			2306	7566	3709	34	758	18154	21		25	84	3.9	4.4	97.7	37.4		17.2	14.1	34	200	5.5
7/8/2015	-247.19	QFP	13954	812	20971	9877	202	1680	44775	10963	25		63	48.3	630	63	89	4.8		4.8	117	2715	16	0.08
7/8/2015	-248.1	QFP	20781		13789	8665	2051	13	735	11058				60	40.4	105	15	74.8		3.4	43	1553	7.9	0.18
7/8/2015	-249.27	QFP	9487		25631	5421	2175	19	915	15117			19	41	4.6	66.6	96.6	11.7		50	1629	277	0.46	
7/8/2015	-250	QFP	20033		31599	7110	1347	16	459	11376	9		44	11.2	80.7	74.5	20.8	79.4		3.3	56	1450	277	0.39
7/8/2015	-250.9	QFP			20414	14985	2476	12	1050	16358				40	4.3	55	121	20.3		51	1339			
7/8/2015	-252	QFP	13954		2306	7566	3709	34	758	18154	21		25	84	3.9	4.4	97.7	37.4		17.2	14.1	34	200	5.5
7/8/2015	-253	QFP	11772		19739	25785	6492	45	1669	39838			92	50.2	157	22.6	81.9	4.3		4.3	59	514	117	0.03
7/8/2015	-253	QFP	3686		24982	19630	3553		1403	27524			12	161	6.3	54.8	123	27.3		2.5	60	855	12.1	0.07
7/8/2015	-254.16	QFP	14405		10591	18822	2279	12	1019	24069			16	270	4	30.3	244	21.9		86.1	49	450	5.7	0.06
7/8/2015	-255	QFP	5136		11903	14767	1630		691	13646			15	53	3.5	32.5	181	13.9		33	608	64.7		
7/8/2015	-256.2	QFP	9780		17102	4671	566	14	578	7044			15	1705	63.1	75.6	143	29.7		35	629	367	0.01	
7/8/2015	-257.3	QFP			13109	18929	2176	17	842	19488	10			71	45.6	175	18.1	84.8		3	37	577	22.4	
7/8/2015	-258.18	QFP	32108	342	12583	22419	5006		1294	32830			261	261	28.4	114	28.3	79.						

Date	Depth (m)	Lithology	S (ppm)	Cl (ppm)	K (ppm)	Ca (ppm)	Ti (ppm)	Cr (ppm)	Mn (ppm)	Fe (ppm)	Ni (ppm)	Cu (ppm)	Zn (ppm)	As (ppm)	Sr (ppm)	Y (ppm)	Zr (ppm)	Nb (ppm)	Ag (ppm)	Ba (ppm)	Pb (ppm)	Cu/Zn		
7/8/2015	-260.63	matrix	6540	446	21974	46238	2885	1180	34022	49	12	47.8	305	17.8	47.8	3.2	47.8	3.2	67	527				
7/8/2015	-261.2	matrix	468	8904	54572	1698	3154	40603	55	11.6	32.1	213	17.3	17.3	77.4	2.8	56	589						
7/8/2015	-262.07	QFP	5440	279	6774	21886	1528	14170	28	7.8	20.1	213	17.3	17.3	77.4	2.8	56	589						
7/8/2015	-263.1	QFP	26369	390	4676	13539	1478	3565	152	13.3	98.8	15.6	48.6	48.6	3.2	53	380	28	40	372				
7/8/2015	-264.25	QFP	34751	576	8930	44720	3565	47242	110	13.2	19.5	220	24.8	24.8	61.9	3.2	53	380	28	53	380	28	0.17	
7/8/2015	-264.9	QFP	4047	617	5500	67597	2190	18273	16	92	11.9	196	19.7	19.7	63.9	2.7	56	197	53	53	445	0.33		
7/8/2015	-266.06	QFP			7515	23399	2202	1131	26938	40	4.5	18.6	207	19.7	67.4	2.7	45	582						
7/8/2015	-267.1	QFP	4859	254	13671	14076	1525	18515	20.8	6.1	41.4	193	20.1	78.5	2.7	45	582							
7/8/2015	-268.2	QFP	60969	700	15767	21447	1904	1528	65225	60	13.7	37	172	21.3	67.5	2.8	85	722						
7/9/2015	-268.95	QFP	2920	4925	84838	1243	337	16023	43	4	13.2	186	17.1	80.7	37	227	37	227						
7/9/2015	-270.3	QFP	9242	4295	84838	446	662	15765	8.9	8.3	400	35.7	24.1	51	109	44	249							
7/9/2015	-271.07	QFP	106535	7340	23709	1635	811	21522	39	3.8	21.3	135	19.2	73.5	2.5	58	362							
7/9/2015	-272	QFP		351	7902	18925	1603	790	18312	33	5.7	20.6	156	15.5	63.7	2.5	58	362					0.22	
7/9/2015	-273	QFP	5705	268	7701	19634	2225	762	20758	11	50	6.7	19.3	170	14.3	60.9	2.7	46	406					
7/9/2015	-274	QFP			19850	29523	2087	2133	33752	94	7.3	61.2	287	25.5	68.9	62	1013							
7/9/2015	-275.1	QFP		298	14500	17759	2192	791	24263	55	7.9	50.4	300	19.8	87.1	3.4	46	294						
7/9/2015	-276.17	QFP			13666	22034	1474	632	19869	40	6.3	40.7	204	14.4	14.4	46	488							
7/9/2015	-277	QFP	1534		17431	31702	2196	876	26803	38	43	7.3	54.8	155	22	75.1	48	386					0.88	
7/9/2015	-278.13	QFP		313	11877	39702	2301	1020	18662	42	6.8	23.8	205	24.6	87.8	3.2	41	394						
7/9/2015	-279.24	QFP	1445		8222	31309	1101	702	18339	46	4.5	24.4	159	17.7	73.5	51	288							
7/9/2015	-280.07	QFP	25891	363	11625	20233	4320	1062	33120	16	1737	13.3	33.7	228	29	109	4.2	46	515	16.5	0.01			
7/9/2015	-280.93	QFP	5705	253	9542	18626	1455	732	15114	10	16	48	4.4	29.1	152	15.5	62.2	47	518				0.33	
7/9/2015	-282.1	QFP		282	10196	23523	2329	1073	23185	15	9	78	6	27.8	176	19.1	80.9	2.9	39	434	6.1	0.12		
7/9/2015	-283.1	QFP			6878	18717	1438	674	18703	25	41	6.9	23.9	190	14.2	71.1	2.5	40	268				0.61	
7/9/2015	-283.94	QFP	73365	512	8555	23295	1981	733	53875	35	43	15.5	28.3	256	18.9	86	3.8	53	411				8.21	
7/9/2015	-285.18	QFP	2039	359	7153	20020	2375	701	17178	22	49	4.7	23.7	184	20.8	83.2	3.2	39	346				0.45	
7/9/2015	-286	QFP	1904		4872	11844	1299	18	634	15215	46	3.8	14.9	148	12.3	58.2	44	242						
7/9/2015	-287.05	matrix		535	9190	56136	3889	62	1345	55752	37	94	14.1	32.2	565	9.8	27.6	5.1	66	281			0.39	
7/9/2015	-288.3	matrix		632	8710	75846	4594	62	1391	53624	17	15	107	14.7	22.1	565	18.1	38.4	2.8	52	176	0.14		
7/9/2015	-289	QFP		318	10751	14665	1440	649	12969	11	34	4	25.2	142	17.6	64.2	2.4	48	337				0.32	
7/9/2015	-290.05	QFP	1194		7665	11528	2112	523	11530	18	41	3.5	24.6	99.3	14	72	35	360					0.44	
7/9/2015	-291	QFP	6895		9550	8712	1355	508	15151	33	4.1	17.8	137	11.6	64.3	38	304							
7/9/2015	-292	QFP	2405	296	5239	12703	1233	583	15683	33	4.1	17.8	137	11.6	64.3	38	304							
7/9/2015	-293.1	QFP	9712	259	9822	20493	1756	793	20538	62	6.2	31.9	260	23.5	78.1	35	544							
7/9/2015	-294.1	QFP	2537	415	8633	26089	2454	989	19186	28	62	5.7	21.6	193	22.6	84.4	41	380					0.45	
7/9/2015	-295.12	QFP	9660	307	4238	17823	1749	697	18864	99	117	6.1	7	151	13.6	60.9	47	485					0.85	
7/9/2015	-296	QFP	2723		9749	104306	1727	1368	32950	46	11	9.5	24.6	348	26.1	47.2	73	841					4.18	
7/9/2015	-297.27	QFP		317	3239	11295	1286	832	20775	38	38	6.7	9.2	234	15.6	75.7	2.4	42	231				3.09	
7/9/2015	-298	QFP			3216	20541	1386	849	17862	170	55	229	16.6	73.6	2.5	42	118	6.7	3.09					
7/9/2015	-299	QFP	10376	309	7383	15280	1632	630	13348	9	295	28.1	106	19.2	75.8	36	415	183	0.03					
7/9/2015	-300	pyritic	7116		319	4593	707	10	184	7825	19	210	4.2	51.3	19.8	118.1	7.2	23	260				0.03	
7/9/2015	-301	pyritic	8700		759	5769	1293	14	588	16979	19	19	757	63.4	27.7	105.6	9.1	41	37	260			0.03	
7/9/2015	-302.26	pyritic	217961	3545	11318	31810	9422	356	4371	252468	359	359	148029	15.7	198	140	117	143	145	405	79	0.00		
7/9/2015	-303	pyritic	51285	1842	12630	56134	4346	107	3784	112337	135	267	12960		84.7	27.5	110.9	5	26	631	11020	0.02		
7/9/2015	-304	pyritic			533	8194	1379	13	591	7734	5	5												
7/9/2015	-305	pyritic			452	3676	1758	17	557	9044	29	29												
7/9/2015	-305.6	pyritic	15300	903	4195	34704	5008	193	3063	80035	43	59	2542		109	29.9	58.7	2.9	75	192	21	0.02		
7/9/2015	-306.1	pyritic	28707	673	7040	18909	6645	216	2137	86297	41	70	304	12.5	129	44	96	8.1	67	537	206	0.23		
7/9/2015	-307	pyritic	26803	713	13956	21200	4875	83	1742	58667	41	31	1760	35.5	150	65	185	12.5	91	1086	909	0.02		
7/9/2015	-308.1	pyritic	43162	658	8149	23851	4870	44	1648	60316	74	40	3098	45	21.5	155	39	177	12.2	83	414	183	0.01	
7/9/2015	-308.9	QFP		349	11841	16792	1918	14	604	13617	37	4.3	38.1	16.1	66.1	2.9	51	843						
7/9/2015	-310.15	QFP	3408		21528	9208	1576	19	429	11352	80	28	28	38.1	174	14.3	73.8	48	1187				2.86	
7/9/2015	-311	QFP	9437	383	9177	12610	2289	28	829	23236	44	44	25.3	182	15.9	70.8	50	321	46					
7/9/2015	-312	QFP		401	22394	22761	2486	98	1239	27222	55	55	66	230	22.7	77.1	2.5	60	872					
7/9/2015	-313	QFP	13124		17539	10916	2203	13	796	21261	29	29	6.5	54.1	123	17.8	76.3	3.2	46	732				
7/9/2015	-314.05	QFP	18206		16040	22556	23	845	22															

Date	Depth (m)	Lithology	S (ppm)	Cl (ppm)	K (ppm)	Ca (ppm)	Ti (ppm)	Cr (ppm)	Mn (ppm)	Fe (ppm)	Ni (ppm)	Cu (ppm)	Zn (ppm)	As (ppm)	Sr (ppm)	Y (ppm)	Zr (ppm)	Nb (ppm)	Ag (ppm)	Ba (ppm)	Pb (ppm)	Cu/Zn	
7/9/2015	-317.35	QFP	208466	1243	10206	15286	1833		2158	150101		451	110	29	29.2	171	52	40.4	109	270	13	4.10	
7/9/2015	-318	QFP	14525	292	9997	22468	2196		984	17139			49	5.4	35.6	136	28.3	60.7	48	371	8.3		
7/9/2015	-319	QFP	2116	585	12866	17007	1829	16	961	17250			41	4.2	31.5	156	24.3	74.4	48	700			
7/9/2015	-320.14	QFP	990	288	6807	6522	2078		690	12738			17.9	3.7	21	129	8.3	63.8	2	40	442		
7/9/2015	-321.13	QFP	50024	536	25067	17567	2276	20	1184	11030		15	25	9.4	54.9	158	20.9	73.6	58	677	6.1	0.60	
7/9/2015	-322	QFP	5282		5375	22750	1269	9	637	11517		13	21.7	15.3	148	16.3	66.4	2.4	36	280		0.60	
7/9/2015	-323	QFP			6118	21552	1472		727	14683			33	4.3	20.1	137	16.8	81.4	2.5	52	341		
7/9/2015	-324	QFP	6162		8234	20717	1345		753	14989			51	5.8	27.3	120	13.9	60.6	40	293			
7/9/2015	-325	QFP		315	7435	18626	1750		810	16568			37	4.1	24.5	153	20.3	71.9	32	466			
7/9/2015	-326	QFP	59447	679	9530	14286	3073		1173	66786			55	14.6	24.5	149	25.8	71.7	4.5	77	656		
7/9/2015	-327	QFP	13598		15393	11647	2085		701	16790			29	5.5	47.2	421	20.4	80.2	47	742			
7/9/2015	-327.85	QFP		306	18492	11416	1113	14	604	8643		43	204	14.3	56.9	133	6.2	46.8	35	1167		0.21	
7/9/2015	-329	QFP	108382	954	11775	21760	2999	19	1311	78136		23	56	14.3	27.8	191	21.2	94	4.1	95	641	0.41	
7/9/2015	-330	QFP	4072	275	7709	12901	1673	14	796	18799			26	3.8	18.1	198	17.7	73.2	46	577			
7/9/2015	-331.1	QFP		324	6993	17519	1508	20	841	18742		138	34	6.9	17.7	214	15.1	72	3.4	46	430	14.7	
7/9/2015	-332	QFP	32232	571	16283	25264	3480	15	1149	31996			25	7.6	41.7	166	29.3	61.2	47	606			
7/9/2015	-333.15	QFP	54324		13028	31525	4501	25	1338	51525		15	54	10.3	32.8	158	25.3	62.2	2.9	61	509		
7/9/2015	-334	QFP	10527	464	10683	41062	1152		665	9941			18.7	12.6	29	69.5	14	30	37	873	48	0.80	
7/9/2015	-335	QFP			13992	15665	2468		917	20578			29	6.9	45.6	221	23.9	67	2.6	49	706		
7/9/2015	-336	QFP			13153	20327	3822	44	1184	30329			60	7.5	37.4	153	22.6	75.3	2.6	51	444		
7/9/2015	-337.22	QFP	22441		24589	13274	3145	16	614	28623		42	35	7.1	74.8	138	21.7	78.4	3.7	51	942	1.20	
7/9/2015	-338.05	QFP		381	23647	13620	2581	31	633	18110			42	4.4	29.4	294	17.4	69.5	2.3	54	434		
7/9/2015	-339.1	QFP	5799	483	16579	86393	2326	32	1205	35695			13	9.9	47.2	351	27.9	58.3	4.3	80	681		
7/9/2015	-340.1	QFP		297	8677	21942	2074	18	786	22378		64	46	7.7	25.6	214	19	80.1	3.4	49	347	1.39	
7/9/2015	-341.1	QFP	14146		20880	25010	2509	13	616	17405			29	11.8	50.3	104	16.6	70.5	2.3	52	1012	6.2	
7/9/2015	-341.89	QFP	1632		13340	116571	1096		1301	35108			20	11.8	37.9	599	30.8	50	4.1	75	453		
7/9/2015	-343	QFP	3982		9074	24341	1601		684	20368			42	4.4	29.4	294	17.4	69.5	2.3	54	434		
7/9/2015	-343.4	matic dyke	1899		5309	75692	5789	279	1715	52545		82	42	12.5	14.4	345	21.9	79	3.8	71	159	0.26	
7/9/2015	-343.7	matic dyke	3260		9849	65143	6671	341	1254	68722		126	128	11	24.8	338	25.2	65.8	74	230			
7/9/2015	-344.1	QFP			3196	958	1913		860	35837		16	71	6.5	17.7	171	10.7	72.9	3.3	33			
7/9/2015	-345	QFP	62564		16962	14019	2522	16	702	37204			48	8.6	51.9	151	15.4	69.2	2.5	63	704		
7/9/2015	-346.2	QFP	98586	702	12698	19572	2268		1215	65843			52	14	32.9	157	25.6	51.5	78	450			
7/9/2015	-347	QFP	3305		5870	21027	1063		661	15309		13	61	22.3	159	19.4	62.6	50	272	50	272	0.21	
7/9/2015	-348.1	QFP	1150	254	9103	16556	1862	16	672	14042			36	36	32.4	177	15.1	81.6	2.8	47	411		
7/9/2015	-348.6	matic dyke	3279		20353	12149	1613	12	539	17947			34	5.1	60.6	158	18.2	66.5	52	931			
7/9/2015	-348.95	matic dyke			9211	81764	6110	177	2054	69928		83	15	104	14.3	21.8	390	23.1	64	5.5	73	360	0.14
7/9/2015	-350	QFP	4083		16122	31367	2194		1318	31867			60	9.3	53.7	273	28.4	35.6	57	401			
7/9/2015	-351.1	QFP	2440		10153	16078	1648	14	873	19320			43	4.4	33.8	431	23.4	68.8	2.9	51	630		
7/9/2015	-352	QFP	12286	316	15048	13234	2530	27	836	20617			55	4.3	45.4	109	17.7	78.4	45	656	5.8		
7/9/2015	-353.1	QFP	4101	249	5539	9474	892		483	10831			23.4	4.2	14	229	11.7	56.6	28	351			
7/9/2015	-353.63	QFP	4850		6817	18426	2002	11	689	21291		34	69	4.9	21.5	174	19.5	71.3	52	357		0.49	
7/9/2015	-353.76	matic dyke		490	3902	76096	5213	432	3894	67289			205	12.2	3.7	134	24.7	56.6	67	125			
7/9/2015	-354.2	QFP	3169		5112	12125	2810		960	21569		297	44	6.2	12.8	359	21.9	71.1	2.7	39	178	6.75	
7/9/2015	-354.9	matic dyke			29132	61426	5259	400	3479	60978			157	12.8	65.2	108	20.9	46.8	2.8	81	1343		
7/9/2015	-355.1	QFP	1742	300	5350	23139	1779	11	578	15941			43	3.9	15.5	167	21.3	75.2	2.9	48	328		
7/9/2015	-356.2	QFP		280	7421	6431	1506		561	17460			50	3.9	15.8	111	19.4	65.6	40	243			
7/9/2015	-356.68	matic dyke			6881	24642	8597	97	1021	68417			120	13.7	13.7	148	22.3	81.2	4.4	63	381		
7/9/2015	-357	matic dyke		590	5970	60547	8208	94	3209	72068			184	13.7	11.5	230	29.7	70.1	3.8	78	307		
7/9/2015	-357.4	matic dyke			8706	38219	8374	95	1939	74659			263	12.2	21.7	202	28.1	73.7	4.5	66	322		
7/9/2015	-358	QFP	1262	408	14619	13761	1922		671	23516		13	40	6.7	47.4	285	13.7	71.9	2.6	58	725	0.33	
7/9/2015	-358.71	matic dyke	6872	755	7059	15559	7015	450	2964	89590			209	15.7	11.9	322	37	66.1	2.9	88	349		
7/9/2015	-359.1	QFP		318	6145	6500	1170		539	16579			31	5.2	12.5	203	17	59.2	36	176			
7/9/2015	-360	QFP		266	5777	12279	1278		566	17427			43	5	23.2	243	17	64.9	45	319			
7/9/2015	-361	QFP	2260		11290	18058	2645	13	782	17487			57	3.9	38.2	100.8	20.8	75.7	42	431			
7/9/2015	-362.05	QFP	4172	528	1843	155994	587	28	1564	46167				12.6	4.5	490	29	20.8	78	203			
7/9/2015	-363.25	QFP	6611		11027	35374	2065	15	1384	38027			54	9.7	26.5	208	25.1	59.8	53	318			
7/9/2015	-364	QFP	8738		10636	24399	1845	14	677	19147		39	43	7.1	28.1	145	23.9	71.4	2.8	53	567	0.91	
7/9/2015	-364.95	QFP	2389		10355	12568	939		401	9977			30		24.6	107	11.4	49.7	32				

Date	Depth (m)	Lithology	S (ppm)	Cl (ppm)	K (ppm)	Ca (ppm)	Ti (ppm)	Cr (ppm)	Mn (ppm)	Fe (ppm)	Ni (ppm)	Cu (ppm)	Zn (ppm)	As (ppm)	Sr (ppm)	Y (ppm)	Zr (ppm)	Nb (ppm)	Ag (ppm)	Ba (ppm)	Pb (ppm)	Cu/Zn		
7/9/2015	-356.2	QFP		280	7421	6431	1506	561	17460	17460	39	50	3.9	15.8	111	19.4	65.6	40	243					
7/9/2015	-356.68	mafic dyke		6881	1021	24642	8597	97	1021	68417	37	120	13.7	13.7	148	22.3	81.2	63	381					
7/9/2015	-357	mafic dyke		590	5970	60547	8208	94	32099	72068	37	184	13.7	11.5	230	29.7	70.1	3.8	78	307				
7/9/2015	-357.4	mafic dyke	1722	408	8706	38219	8374	95	1939	74659	56	40	6.2	21.7	202	28.1	73.7	4.5	66	322			0.33	
7/9/2015	-358	QFP	1262	408	14619	13761	9322	671	23516	13761	56	13	13.7	47.4	285	13.7	71.9	2.6	58	725				
7/9/2015	-358.71	mafic dyke	6872	755	7059	18559	7015	450	2964	89590	56	209	15.7	11.9	322	37	66.1	2.9	88	349				
7/9/2015	-359.1	QFP		266	5777	12279	1170	559	16379	16379	56	31	5.2	12.5	203	17	59.2	3.6	176	176				
7/9/2015	-360	QFP		266	5777	12279	1170	559	16379	16379	56	31	5.2	12.5	203	17	59.2	3.6	176	176				
7/9/2015	-361	QFP	2260	611	11290	18058	2645	13	782	17487	2645	57	38.2	100.8	20	64.9	45	319	43	319				
7/9/2015	-362.05	QFP	4172	528	1843	155994	587	28	1564	46167	587	54	12.6	4.5	490	29	20.8	78	203	48				
7/9/2015	-363.25	QFP	6611	11027	35374	2065	15	1384	38027	38027	1845	39	43	7.1	28.1	145	23.9	71.4	2.8	53	318		0.91	
7/9/2015	-364	QFP	8738	10636	24399	1845	14	401	9977	9977	1845	30	24.6	107	11.4	49.7	32	451	32	451				
7/9/2015	-364.95	QFP	2389	10355	12568	939	401	9977	9977	939	401	9977	30	24.6	107	11.4	49.7	32	451	32	451			
7/9/2015	-365.53	QFP	53664	311	3689	25605	548	983	30694	30694	98	50	8.8	8.6	91	25.2	33.3	43	194	43	194		1.96	
7/9/2015	-365.65	QFP	199095	812	864	93735	131	662	107491	107491	535	16	30	8.8	8.6	91	25.2	33.3	43	194	43	194	33.44	
7/9/2015	-367.15	QFP	2321	2831	31668	1587	131	662	107491	107491	535	16	30	8.8	8.6	91	25.2	33.3	43	194	43	194	33.44	
7/9/2015	-368.95	QFP	5769	339	3899	19566	2564	13	653	15706	15	29	3.9	6.3	207	23.3	90.4	38	128	47			0.52	
7/9/2015	-368.95	QFP	11998	472	10070	44875	7860	49	2392	74384	25	51	16.2	14.7	43.8	22.1	25.6	10.3	73	368			0.38	
7/9/2015	-370	mafic dyke	2969	488	19912	6972	88	1927	57138	57138	25	51	13.3	14.7	43.8	22.1	25.6	10.3	73	368			0.38	
7/9/2015	-370.36	chert	5901	543	6301	5953	2919	106	1649	57818	24	152	9.7	15.8	118	8.4	15.9	5.7	366				0.69	
7/9/2015	-371.1	chert	894	373	3617	823	627	12	556	11338	19	10	14.5	5.5	23.1	28	45.8	2.9	45	45			0.13	
7/9/2015	-372.56	chert	15941	1138	1761	10767	2318	17	763	24101	34	24	7.6	11.3	12.2	12.1	59.2	2.7	318				0.13	
7/9/2015	-372.88	chert		647	5237	6305	1365	408	14667	14667	62	27	6.1	9.8	69.5	26.7	120	5	41	93	18			
7/9/2015	-374	mafic dyke		647	5237	6305	1365	408	14667	14667	62	27	6.1	9.8	69.5	26.7	120	5	41	93	18			
7/9/2015	-374.53	mafic dyke		819	13027	82757	5093	165	2033	52003	76	124	9.8	12.2	352	19.6	70	6.1	72	373				
7/9/2015	-375.1	chert	786	418	2576	2498	845	40	107	5756	65	75	11.7	32.7	263	21.4	67.4	4.3	78	1177				
7/9/2015	-376.1	chert	76471	468	8609	12193	1714	40	296	64308	50	34	25	22	12.2	39.4	12.5	85.9	4.5	21	308			
7/9/2015	-377.8	chert	15207	682	687	8283	3198	22	607	39420	56	20	7.6	11.3	11.4	15.0	33.9	83	4.9	76	392		23	
7/9/2015	-378.8	chert	28301	541	3474	179090	2842	32	2957	107193	40	17	17	27	12.2	1649	10	20	6.9	97			13.6	
7/9/2015	-379	mafic dyke	3166	919	3675	72069	9443	32	2957	107193	40	17	17	27	12.2	1649	10	20	6.9	97			0.26	
7/9/2015	-380	mafic dyke	4101	797	7553	42009	9403	166	2438	102588	40	32	16.1	14.1	17	379	31.3	58	80	531			0.35	
7/9/2015	-381	mafic dyke		401.5	12251	45432	4013	166	2438	102588	40	32	16.1	14.1	17	379	31.3	58	80	531			0.20	
7/9/2015	-382.7	chert	8313	398	1761	10767	2318	17	763	24101	34	24	7.6	11.3	11.4	15.0	33.9	83	4.9	76	392		0.19	
7/9/2015	-382.9	chert	165184	1138	1761	10767	2318	17	763	24101	34	24	7.6	11.3	11.4	15.0	33.9	83	4.9	76	392		0.19	
7/9/2015	-384	chert	3807	399	2314	5738	1431	19	479	121947	33	29	8.41	20	9.1	145	31.6	79	86	526	10		0.09	
7/9/2015	-384.9	chert	83790	590	3318	3405	2925	24	692	56551	54	29	14.6	16.8	46.7	54.6	12	35.7	5.2	76	43		0.27	
7/9/2015	-386	chert	18453	2934	10528	29264	2591	37	1049	33454	44	321	10	33.4	159	53	121	7.2	57	342	44		4.22	
7/9/2015	-386.9	chert	4363	290	2468	6941	1617	482	11421	11421	13	44	16.2	6.6	5.2	8.1	89.5	5.4	83	74	18.5		0.33	
7/9/2015	-388.15	mafic dyke	69131	877	1570	43485	4389	140	2484	123330	46	162	19	24.4	29	55	5.4	83	74	18.5			0.24	
7/9/2015	-389	mafic dyke	101801	1204	3795	64932	1654	41	1328	175944	24	18	16.8	16.8	46.7	54.6	12	35.7	5.2	76	43		0.33	
7/9/2015	-390.15	mafic dyke	7873	566	15358	141307	2876	38	952	45687	28	32	21	5.8	11.8	20.2	99	5.2	63	231.2			0.33	
7/9/2015	-391.15	cherty pyritic	67823	743	3420	1484	2717	645	645	36117	31	10	41	8.9	6.6	18.2	18	104.9	6.5	54			0.24	
7/9/2015	-393.3	cherty pyritic	20434	719	1931	975	2052	18	569	8685	21	46	33	20.6	55	30.2	93.9	6.9	23	62			0.24	
7/9/2015	-394.2	cherty pyritic	217397	1009	7632	9804	1186	37	873	109599	59	59	33	20.6	55	30.2	93.9	6.9	23	62			0.24	
7/9/2015	-396	cherty pyritic		961	757	163862	2669	37	873	109599	59	59	33	20.6	55	30.2	93.9	6.9	23	62			2.19	
7/9/2015	-397	cherty pyritic	10152	511	1709	20723	3595	27	1311	67947	23	50	14.8	52.4	10.8	25.7	82	86	68	319	45		0.46	
7/9/2015	-398	cherty pyritic	88235	720	3639	26742	3073	42	1928	72854	181	240	9.6	6.8	133	13.9	47.5	2.5	46	68			0.75	
7/9/2015	-398.95	cherty ash		351	7844	14274	3468	28	1449	24081	24	17	13.7	23.8	18.2	51.7	3.4	3.4	71	155			9.09	
7/9/2015	-399.0	cherty ash	97392	810	5691	15343	2026	14	490	16886	35	156	14.9	16.3	396	10.4	9.9	13.5	40	66			0.36	
7/9/2015	-400.1	cherty ash	8190	749	2521	46637	2172	17	490	16886	35	156	14.9	16.3	396	10.4	9.9	13.5	40	66			0.36	
7/9/2015	-402	mafic		509	10026	73149	557	54	1481	76921	20	141	7.2	11.4	24	25.7	6.1	15.9	79	285			1.05	
7/9/2015	-402.95	mafic	11914	959	30444	43932	2519	80	2314	99301	14	107	14.4	16.8	189	7.2	73.3	203	8	106			1.96	
7/9/2015	-404.07	mafic		1158	26041	27505	3009	64	749	69339	14	107	14.4	16.8	189	7.2	73.3							

Appendix C

Summary of microthermometric data.

Mineral	ID	Eutectic (°C)	Hydrohalite melting (°C)	CaCl ₂ -hydrate melting (°C)	Vapor Homogenization (°C)	Halite Dissolution (°C)	CaCl ₂ equivalent	CaCl ₂ (wt%)	NaCl (wt%)	S _{wt}	Density (g/cc)	Phases
Cal B	106.17			3.1	118.9							L + V
Cal B	106.17				236.3	176.1						L + V + H
Cal B	106.17			31.7		205.0		43.5	10.0	53.5		L + V + H
Cal B	106.17			31.7	164.2	204.8		43.6	9.9	53.5	1.51	L + V + H
Cal B	106.17											
Cal B	106.17	-47.4	-16.6		88.6	greater than 304						L + V + H
Cal B	106.17	0.3	-1.9		223.8							L + V
Cal B	106.17	-50.2			157.2							L + V
Cal B	106.17	-37.5	-22.6		289.8							L + V
Cal B	106.17	-37.5	-20.7									L + V
Cal B	106.17				186.3	greater than 240						L + V + H
Cal B	106.17		-8.2		172.5							L + V
Cal B	106.17		-7.8		111.1							L + V
Cal B	106.17		-8.2		119.6							L + V
Cal B	106.17				142.8							L + V
Cal B	106.17				152.7							L + V
Cal B	106.17				206.4							L + V
Cal B	106.17		-23.7	35.4	227.2			49.8	0.1	49.8	1.67	L + V
Cal B	106.17		-24.0		224.1							L + V
Cal B	106.17	-38.6	-23.4		212.2	greater than 280						L + V + H
Cal B	106.17		-23.4		181.0							L + V
Cal B	106.17	-63.2			240.2							L + V
Cal B	106.17				261.4							L + V
Cal B	106.17			12.2	203.9		40				1.38	L + V + H
Cal B	106.17				190.9							L + V
Cal B	106.17				194.6							L + V
Cal B	106.17	-40.3	-18.1		171.3							L + V
Cal B	106.17	-65.7	-30.7		197.9							L + V
Cal B	106.17	-65.8	-21.9		296.2							L + V
Cal B	106.17				175.2	200.5						L + V + H
Cal B	106.17	-65.7	-30.1		209.6							L + V
Cal B	106.17				176.9							L + V
Cal B	106.17				179.2							L + V
Cal B	106.17	-52.7	-21.1		205.5							L + V
Cal B	106.17				181.8							L + V
Cal B	106.17				172.4							L + V
Cal B	106.17	-60.2	-15.2		231.3							L + V
Cal B	106.17				226.5							L + V + H
Cal B	106.17	-51.8	-37.6	85.9	233.0		59				2.69	L + V
Cal B	106.17				261.6							L + V
Cal B	106.17				252.7							L + V
Cal B	106.17				237.4							L + V
Cal B	106.17			28.0	178.8		47				1.56	L + V
Cal B	106.17			28.4	164.9	209.8		42.2	10.8	53.0	1.47	L + V + H
Cal B	106.17		-12.7	28.4	186.8	214.8		42.0	11.3	53.3	1.46	L + V + H
Cal B	106.17	-46.0	-12.7	28.4	200.1		47				1.54	L + V
Cal B	106.17		-12.2	28.4	186.1		47				1.55	L + V
Cal B	106.17	-81.6	-32.5	21.2	220.0			42.2	0.7	42.9	1.37	L + V
Cal B	106.17		-32.5	21.2	232.5			42.2	0.7	42.9	1.37	L + V
Cal B	106.17	-52.4		31.1	235.2		50				1.65	L + V
Cal B	106.17	-52.4		17.3	162.2		42				1.40	L + V
Cal B	106.17	-52.4		17.9	165.5		42				1.40	L + V
Cal B	106.17	-53.2		19.9	179.8		43				1.42	L + V
Cal B	106.17				220.1							L + V
Cal B	106.17				164.0							V + L
Cal B	106.17				199.8							V + L
Cal B	106.17				220.7							V + L
Cal B	106.17				267.5							V + L

Cal B	106.17	-57.1	-34.4		296.2						V + L
Cal B	106.17	-64.0	-52.0	22.3	223.0		42.8	0.7	43.4	1.39	V + L
Cal B	106.17	-64.0	-52.2	22.0	198.2		42.6	0.7	43.3	1.40	V + L
Cal B	106.17	-64.0	-52.0	21.5	178.9		42.2	0.7	43.0	1.40	V + L
Cal B	106.17										
Cal B	106.17	-51.4		22.9	104.9	44				1.50	V + L
Cal B	106.17			15.5	111.5	41				1.41	V + L
Cal B	106.17	-72.1	-38.2		238.0						V + L
Cal B	106.17			21.6	235.7	44				1.39	V + L + H
Cal B	106.17			25.2	196.6	46				1.49	V + L
Cal B	106.17			18.2		274.6	32.5	20.0	52.5		V + L + H
Cal B	106.17			24.5	232.1	45				1.43	
Cal B	106.17			24.5	233.0	45				1.45	
Cal B	106.17			24.5	304.4	45				1.39	
Cal B	95.10				186.9						V + L
Cal B	95.10				188.2	212.7					V + L + H
Cal B	95.10				184.4						V + L
Cal B	95.10				191.2						V + L
Cal B	95.10				119.0						V + L
Cal B	95.10										L + H
Cal B	95.10	-25.6	-14.7		201.4						V + L
Cal B	95.10				191.6						V + L
Cal B	95.10										L + H
Cal B	95.10										L + H
Cal B	95.10	-32.5		26.4	208.8	46				0.85	V + L + H
Cal B	95.10			31.5							V + L + H
Cal A	95.10					212.0					L + H
Cal A	95.10					197.3					L + H
Cal A	95.10		-12.7	76.1	209.6	58				2.52	V + L + H
Cal A	95.10				218.8						V + L
Cal A	95.10				227.7						V + L
Cal A	95.10				255.7						V + L
Cal A	95.10	-34.0		183.1	266.7						V + L
Cal A	95.10	-34.0			271.5						V + L
Cal A	95.10	-34.0			286.3						V + L
Cal A	95.10	-34.0			276.3						V + L
Cal A	95.10	-34.0			233.3						V + L
Cal A	95.10				283.9						V + L
Cal A	95.10	-41.8	-22.1		248.3						V + L
Cal A	95.10	-41.5		28.4	295.5	47				1.49	V + L
Cal A	95.10		-13.0		314.0						V + L
Cal A	95.10			26.4							V + L
Cal A	95.10			26.4	229.0	46				1.48	V + L
Cal A	95.10	-54.5	-30.6		269.8						V + L
Cal A	95.10	-54.5	-28.6		314.2						V + L
Cal A	95.10				327.0						V + L
Cal A	95.10				267.8						V + L
Cal B	200.20				74.4						V + L
Cal B	200.20				221.9						V + L
Cal B	200.20				165.7						V + L
Cal B	200.20			25.9	221.5	46				1.47	V + L
Cal B	200.20			26.1	253.0	46				1.47	V + L
Cal B	200.20			26.1							V + L
Cal B	200.20				225.0						V + L
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Averages		-48.8	-23.8	31.3	210.9	210.8	46	42.3	6.0	48.3	1.51